

**PSG INSTITUTE OF TECHNOLOGY AND APPLIED RESEARCH**

**COIMBATORE – 641 062**

(Autonomous college affiliated to Anna University)



**2025 Regulations**

**Regulations and Courses of Study,  
Scheme of Assessment and  
Syllabi for First and Second Semesters**

**for**

**B.E. and B.Tech. Degree Programmes**

**PSG INSTITUTE OF TECHNOLOGY AND APPLIED RESEARCH**  
**COIMBATORE – 641 062**  
(Autonomous college affiliated to Anna University)

**2025 REGULATIONS FOR FOUR-YEAR B.E. AND B.Tech. DEGREE PROGRAMMES**  
(for students admitted in academic year 2025 – 2026, and subsequently under Choice-Based Credit System) \*

**NOTE:** The regulations hereunder are subject to amendments as may be made by the academic council of the college from time to time. Any or all such amendments will be effective from such date, and to such batches of students (including those already undergoing the programme) as may be decided by the academic council.

**1. PRELIMINARY DEFINITIONS AND NOMENCLATURE**

**1.1.** In the following regulations unless the context otherwise requires:

**1.1.1. Programme** means B.E./B.Tech. degree programme

**1.1.2. Branch** means specialisation or discipline of B.E./B.Tech. degree programmes, like Artificial Intelligence and Data Science, Civil Engineering, etc.

**1.1.3. Course** means a theory or laboratory-integrated theory or practical course that is normally studied in the semester

**1.1.4. University** means Anna University, Chennai

**1.2. CONDITIONS FOR ADMISSION**

Students for admission to the B.E./B.Tech. degree programme will be required to satisfy the conditions of admission prescribed by the University and Government of Tamilnadu.

**2. DURATION OF THE PROGRAMME\***

**2.1. Minimum Duration:** The programme will extend over a period of four years leading to the Degree of Bachelor Engineering (B.E.) / Bachelor of Technology (B.Tech.) of Anna University. The four academic years will be divided into eight semesters with two semesters per academic year. Each semester shall normally consist of 90 working days including examination days.

**2.2. Maximum Duration:** The student shall complete all the passing requirements of the B.E./B.Tech. degree programme within a maximum period of 7 years (6 years for lateral entry); these periods are reckoned from the commencement of the semester to which the student was first admitted to the programme.

*\* Provision is made for lateral entry of students in the third semester of the programme in all branches of study and they will be required to satisfy the conditions of admissions thereto prescribed by the Anna University and Government of Tamil Nadu.*

### 3. B.E./B.Tech. PROGRAMMES OFFERED

The following are the branches of study under B.E./B.Tech. degree programme:

- BE Civil Engineering
- BE Computer Science and Engineering
- BE Electrical and Electronics Engineering
- BE Electronics and Communication Engineering
- BE Instrumentation and Control Engineering
- BE Mechanical Engineering
- B.Tech. Artificial Intelligence and Data Science
- B.Tech. Electronics Engineering (VLSI Design and Technology)

### 4. STRUCTURE OF PROGRAMMES

**4.1. Course Work:** The course work of the odd semesters will normally be conducted in odd semesters and that of even semesters, in even semesters.

**4.2. Curriculum:** The curriculum will comprise courses of study as given in *section 17* in accordance with the prescribed syllabi. The hours per week mentioned for each of the courses refers to periods per week. The curriculum consists of:

- 4.2.1.** Basic Sciences (BS) courses include Mathematics, Physics, Chemistry, Biology, etc.
- 4.2.2.** Humanities, Social Sciences and Management Courses (HSMC) include Professional English, Communication skills, etc.
- 4.2.3.** Engineering Sciences (ES) courses include Engineering Practices, Engineering Graphics, Basics of Civil / Mechanical / Electrical / Electronics / Instrumentation / Computer Engineering, etc.
- 4.2.4.** Professional Core (PC) courses include the core courses relevant to the chosen branch.
- 4.2.5.** Professional Elective (PE) courses include the elective courses relevant to the chosen branch. Professional elective courses are offered under verticals (specialisation groups)
- 4.2.6.** Open Elective (OE) courses include the courses offered by a branch to other branches, from the list specified in the respective curriculum of the B.E. / B. Tech. programmes
- 4.2.7.** Employability Enhancement Courses (EEC) include mini-project, project work, internship, soft skills development / business and managerial communications / quantitative and reasoning skills/ problem solving courses, etc.
- 4.2.8.** Mandatory Courses (MC)
- 4.2.9.** Induction Programme (IP) and
- 4.2.10.** AICTE Activity Point Programme

**4.3. Electives:** Every student shall opt for electives from the list of electives of the respective degree programme as given in *section 17*, in consultation with tutor, programme coordinator and the head of the department. A student shall study six professional elective courses and two open elective courses. The professional elective and open elective courses will be offered from the 4<sup>th</sup> semester to the 8<sup>th</sup> semester. The minimum number of credits to be earned for professional elective courses is 18 and the minimum number of credits to be earned for open elective courses is 6. The professional elective courses are offered by the department to the students of their own branch only, and open elective courses are courses offered by a department to students of other branches.

Students can opt out of a maximum of three elective courses (both professional and open elective courses put together), by studying online courses, self-directed learning courses and/or three one-credit courses (equivalent to one elective course). Only one elective course can be opted out in lieu of one-credit courses

**4.4. Online Course (SWAYAM based NPTEL):** Students can register and earn credits for NPTEL courses approved by the department committee consisting of the head of the department, the programme coordinator, and a faculty member who is a subject expert. The list of NPTEL courses is to be approved by the academic council on the recommendation of the chairperson of the respective board of studies before the beginning of the corresponding semester, or may be ratified in the subsequent academic council meeting, as approved by the head of the institution.

Students may pursue NPTEL courses from 4<sup>th</sup> semester to 7<sup>th</sup> semester for getting exemption for professional/open elective courses in the subsequent semester. Any NPTEL course studied by a student during break of study period shall not be considered for exempting a professional elective or an open elective course.

The NPTEL courses of duration 12 weeks will be considered as 3-credit courses. The mapping of the marks with the grades is given in Table 1.

**Table 1 Grade Assignment based on NPTEL Mark**

Letter Grade	Mark Range
O	90 - 100
A+	80 - 89
A	70 - 79
B+	60 - 69
B	50 - 59
C	40 - 49

**4.5. Self-Directed Learning Course:** These courses are not text book-based courses but are based on various attempts made by researchers to solve a specific problem as available in quality journals. These courses will not have specific course titles. The course under this category will have three credits.

A student, under the guidance of a faculty member who has completed a PhD degree, can register for the self-directed learning course from the 3<sup>rd</sup> semester, after obtaining approval from the head of the institution for the recommendation made by the department committee consisting of the head of the department, programme coordinator, and a faculty member who is a subject expert. The student should choose a specific problem after conducting a literature survey and solve the problem. An article based on this work must be communicated for publication in journals that are indexed in both SCOPUS and/or SCI databases. The respective student and the guide shall be the first and second authors, and the maximum number of authors permitted is two. After the publication of the article, the student can seek exemption for one professional/open elective course in the subsequent semester (from the 5<sup>th</sup> semester to the 8<sup>th</sup> semester for B.E. and B.Tech. programmes. The grade will be awarded based on the quality of publication, as mentioned in Table 2.

**Table 2 Grade Point Assignment for Self-Directed Learning Course**

Category of Journal	Grade Point
Publication that is indexed in both SCOPUS and SCI databases	O
Publication that is indexed in SCOPUS or SCI database	A+

**4.6. Project Work:** Every student shall be required to undertake a suitable project in industry / research organisation / department in consultation with the head of the department and the faculty member guide, and submit the project report thereon at the end of the semester in which the student registered for the project work on date announced by the college/department. For B.E. and B. Tech. programmes students shall register for the mini-project I in 4<sup>th</sup> semester, mini-project II in the 6<sup>th</sup> semester, project work I in 7<sup>th</sup> semester and project work II in 8<sup>th</sup> semester.

**4.7. Internship / Community project:** For B.E. and B.Tech. programmes, every student has to earn credit by internship / community project in the 5<sup>th</sup> semester. Internship I / community project is to be carried out during the 4<sup>th</sup> semester vacation. The community project must provide a sustainable solution for a socially relevant problem. Additionally, every student shall earn one credit by internship in the 7<sup>th</sup> semester, by taking up Internship II during the 6<sup>th</sup> semester vacation. An internship / community project of one credit is equivalent to 40 hours of work.

**4.8. Induction Programme:** All students shall undergo an induction programme at the beginning of the first semester for a duration of three weeks as per the guidelines of All India Council for Technical Education (AICTE). A student completing the induction programme will be awarded a completed grade in the grade sheet, and only the students who complete the induction programme shall be considered as eligible for award of degree subject to satisfying other conditions. A student who does not complete the induction programme in the first semester shall redo the same in the subsequent semester.

**4.9. AICTE Activity Point Programme:** Every student is required to earn activity points by performing community service and allied activities, in addition to earning the required academic credits. The regular student, who is admitted to the 4-year degree programme, is required to earn a minimum of 100 activity points. The student entering the 4-year degree programme through lateral entry is required to earn a minimum of 80 activity points. Three hours of participation is equivalent to one activity point. From the 2<sup>nd</sup> semester (3<sup>rd</sup> semester for lateral entry) to the 6<sup>th</sup> semester, 20 activity points are to be earned every semester, by performing the activities during the weekends and/or holidays during the semester. These activities will be coordinated by NSS / NCC / Sports / Saansad Adarsh Gram Yojana (SAGY) coordinator / career development centre (CDC) and/or the student department.

Every student is required to prepare a file containing documentary proof of activities, done by him/ her. This file will be duly verified by NSS / NCC / Sports / Saansad Adarsh Gram Yojana (SAGY) coordinator / career development centre / department. The student will be provided a certificate from the concerned coordinator and head of the institution.

The student should earn all the specified activity points before he/she appears for his/her end semester examinations. The activity points students have earned will be mentioned on the student's transcript. However, there will be neither grades/marks for these activity points (these points will not be accounted for CGPA calculation). If a student fails to earn all the prescribed activity points, the 8<sup>th</sup> semester grade sheet shall be issued only after earning those activity points.

**4.10. One-Credit Course:** Students can also opt for one-credit industry-oriented courses, each of 15 hours duration for theory course, and 30 hours for practical course, offered by experts from industry / other institutions on specialized topics. Students can complete such one-credit courses during the 3<sup>rd</sup> semester to 7<sup>th</sup> semester, as and when these courses are offered by the departments. The student may also register for the one-credit courses offered by departments other than the student department, provided the student has fulfilled the necessary prerequisites of the course as prescribed by the department offering the course.

There is no limit on the number of one-credit courses a student can register and successfully complete. However, the number of professional/open elective courses for which the student seeks exemption for having studied one-credit courses shall not exceed one.

After completing the required number of one-credit courses, if a student wishes to avail exemption for a professional elective course, he/she can do so in writing to the respective head of the department, before the commencement of 5<sup>th</sup> or 6<sup>th</sup> or 7<sup>th</sup> or 8<sup>th</sup> semester. The grades of the students successfully completing the one-credit courses will be based on the absolute grading system.

The grades earned by the students for the one-credit courses which are not opted for conversion in lieu of a professional/open elective course, will not be included in the computation of CGPA.

**4.11. Mandatory Courses:** The student shall study the mandatory courses prescribed and the grade will be mentioned in the grade sheet. However, it will not be considered for computation of CGPA.

**4.12. Course Enrollment and Registration:**

- 4.12.1. Each student, on admission shall be assigned to a faculty mentor for the entire duration of the programme. During enrolment, the faculty mentor shall advise and counsel the student about the details of the academic programme and the choice of courses considering the student's academic background and career objectives.
- 4.12.2. Each student on admission shall register for all the courses prescribed in the curriculum in the student's first semester of study. In the case of lateral entry students, they shall register for all the courses prescribed in the curriculum in the third semester of study.
- 4.12.3. From third semester onwards (fourth semester in the case of lateral entry students), a student has the option to drop one theory course (except professional core courses) in a semester. The maximum number of credits the student can register in a particular semester cannot exceed 36 credits including courses registered for Honours / Minor Degree, and the course for which the student has registered for redo.
- 4.12.4. In case of a student dropping a course of study in any semester, he/she shall register for that course in the next given opportunity and also earn necessary attendance in that course to become eligible to appear for the end-semester examination in that course.
- 4.12.5. The courses to be offered in any semester for candidates who need to reappear or have attendance shortage, etc. will be decided by the head of the department.

**4.12.6.** After registering for a course, a student shall attend classes, satisfy the attendance requirements, earn continuous assessment marks, and appear for the end-semester examinations.

The enrolment for all the courses of the 2<sup>nd</sup> semester will commence 10 working days prior to the last working day of the 1<sup>st</sup> semester. The student shall confirm the enrollment by registering for the courses within the first 5 working days after the commencement of the 2<sup>nd</sup> semester. The enrollment for the courses of the 3<sup>rd</sup> semester to 8<sup>th</sup> semester will commence 10 working days prior to the last working day of the preceding semester.

The student shall enroll for the courses with the guidance of the faculty mentor. If the student wishes, the student may drop a course within 5 working days after the commencement of the corresponding semester and complete the registration process duly authorised by the faculty mentor.

**4.13. Credit Assignment:** Each course is assigned a certain number of credits based on Table 3.

**Table 3 Contact Period per Week and Credits**

Contact Period per Week	Number of Credits
One lecture period	1
One tutorial period	1
Two practical periods (laboratory course, project work, etc.)	1

**4.14. Minimum Credits:** The minimum number of credits to be earned through successful completion of the courses of study in the respective branch by a student to qualify for the award of degree, is provided in Table 4.

**Table 4 Minimum Credits to be Earned for the Award of Degree**

Programme	Minimum Number of Credits to be Earned for the Award of Degree	
	For entry at First Semester	For Lateral Entry at Third Semester
B.E. Civil Engineering	168	123
B.E. Computer Science and Engineering	168	122
B.E. Electrical and Electronics Engineering	168	120
B.E. Electronics and Communication Engineering	168	119
B.E. Instrumentation and Control Engineering	168	122
B.E. Mechanical Engineering	168	122
B.Tech. Artificial Intelligence and Data Science	168	120
B.Tech. Electronics Engineering (VLSI Design and Technology)	168	119

Maximum number of credits is 172 and 128 for regular and lateral entry students respectively

**4.15. Entrepreneurship and Innovation Startups:** Student entrepreneurs may avail exemption from joining the courses at the beginning of 5<sup>th</sup>, 6<sup>th</sup> and 7<sup>th</sup> semesters for a maximum duration of 15 days in each of these semesters. This exemption period can be used for entrepreneurship related activities like attending entrepreneurship / innovation training programmes conducted by premier institutes, working with startups etc. The exemption can be availed by getting prior approval before the commencement of the respective semester from a committee constituted by the head of the department and approved by the head of institution. The students should demonstrate how productively their exemption period was utilised, through a presentation/viva voce to the committee, after the end of the exemption period in the respective semester.

**4.16. Medium of instruction:** English is the medium of instruction for examinations, project reports, etc. other than elective language courses.

## **5. REQUIREMENTS OF ATTENDANCE AND PROGRESS**

**5.1.** A student will be qualified to appear for end-semester examination in a particular course of a semester only if he/she has satisfied the attendance requirements as per the norms given below:

**5.1.1.1.** Shall secure not less than 75% attendance in that course

**5.1.1.2.** If a student secures attendance 65% or more but less than 75% in any course in the current semester due to medical reasons (hospitalisation/accident/specific illness) or due to participation in the college/university/state/national/international level sports events with permission from the head of the institution, the student shall be given exemption from the prescribed attendance requirement and the student shall be permitted to appear for the end-semester examinations of that course.

5.1.2. His/her progress has been satisfactory and

5.1.3. His/her conduct has been satisfactory.

**5.2.** A student shall normally be permitted to appear for end-semester examinations of the course if the student has satisfied the attendance requirements and has registered for examination in those courses of that semester by paying the prescribed fee.

**5.2.1.** Students who do not satisfy sections under section 5.1 will not be permitted to appear for the end-semester examinations / evaluation of that course. The student has to register and redo that course in a subsequent semester when it is offered next, earn necessary attendance and continuous assessment marks and appear for end-semester examinations.

**5.2.2.** If the total number of redo courses (except mandatory courses) at the end of any semester is more than two, the student will not be eligible to register for next immediate semester courses. Such students will be permitted to register for those courses only when offered next, subject to fulfilment of the condition on the maximum number of redo courses.

- 5.2.3.** If a student with more than two redo courses (except mandatory courses) is in the last batch of his/her current regulations, then he/she has to redo two equivalent courses from the next regulations when it is offered. If such equivalent courses are not available in the next regulations, he/she has to complete the redo courses in a self-study mode, by getting prior approval from the head of the department who will nominate a faculty member for the periodic monitoring and evaluation of the course.
- 5.2.4.** A student who has already appeared for a course in a semester and passed the examination is not entitled to reappear in the same course for improvement of grades / marks.
- 5.2.5.** In respect of students who complete a part of the academic programme either one or two semesters under the student exchange scheme in approved foreign universities, the transfer of credits of equivalent courses successfully completed by them in the foreign university will be approved. For the remaining courses of the respective semester(s) which they have not studied in the applicable regulation, they shall register for those courses within the next two or subsequent semesters on a self-study basis, and the scheme of evaluation applicable for those courses will be followed. Such an appearance of the student in those courses will be treated as the first appearance for the purpose of classification.

## **6. DISCIPLINE**

- 6.1.** Every student is required to observe discipline and decorum both inside and outside the college and not indulge in any activity which will bring down the reputation of the college. The head of the institution shall constitute a disciplinary committee to enquire into acts of indiscipline and notify the punishment.
- 6.2.** If a student indulges in malpractice in any of the examinations, he/she shall be liable for punitive action as decided by the head of the institution.

## **7. PROCEDURE FOR REJOINING THE PROGRAMME**

A student who desires to rejoin the programme after a period of discontinuation or who upon his/her own request is permitted by the authorities to study, may join the semester which he/she is eligible or permitted to join, only at the time of its normal commencement for a regular batch of students, and after obtaining the approval from the Commissioner of Technical Education and the University. No student will however be enrolled in more than one semester at any time.

## **8. ASSESSMENT AND PASSING REQUIREMENTS**

- 8.1. Assessment:** The assessment will comprise of end-semester examination (ESE) and/or continuous assessment (CA), carrying marks as specified in the scheme in *section 17*. The continuous assessment marks will be awarded by assessing the student continuously during the semester. The assessment for all the theory courses will be done on the relative grading system. If the student's strength is less than or equal to 30 for a particular theory course, they will be assessed by the absolute grading

system. For the purpose of reporting the performance of a student, letter grades and grade points will be awarded as specified in *section 8.3*.

**8.2. End-Semester Examinations:** End-semester examinations will normally be conducted during October/November and during March/April of each year. Reappearance examinations may be conducted at such times as may be decided by the college. A student will be permitted to appear for the end-semester examination in a course, only if he/she has completed the study of that course.

**8.3. Grade and Grade Point:** Each student, based on his/her performance, will be awarded a letter grade and grade point as specified in *sections 8.3.1 to 8.3.3* for each course at the end of each semester by following the relative grading system and/or absolute grading system.

**8.3.1. Relative Grading System:** In this system, the grades are awarded to the students based on their performance relative to other students in theory courses having continuous assessment and end-semester examination components. For each theory course, the total mark M (i.e. the sum of continuous assessment marks and end-semester examination marks) is computed for every candidate. In theory courses, if a student secures a mark as detailed below he/she declared as fail (RA) in that theory course.

Marks scored in ESE is less than 45% (or) M less than 50% of total marks	Grade: RA
--	-----------

RA denotes reappearance in a course

**Note:** After omitting the marks (M) of all failed students, if the number of students who have passed the course is more than 30, the marks obtained by all the students in that course (having the same course code) will be normalised using the BOX-COX transformation method. The grade range for each course will be computed based on the procedure given by Anna University, Chennai and the grades will be awarded for each student in a particular course. This procedure will be followed for all the applicable courses offered under this regulation. The letter grade and grade point to each student are awarded as given in Table 5.

**Table 5 Relative Grade and Grade Point**

Letter Grade	Grade Point (g)
O (Outstanding)	10
A+ (Excellent)	9
A (Very Good)	8
B+ (Good)	7
B (Average)	6
C (Satisfactory)	5
RA (Re-appearance)	0
SA (Shortage of attendance)	0
W (Withdrawal)	0

**8.3.2. Absolute Grading System:** If the number of students registered for a particular course or if the number of students who have passed a particular course is less than or equal to 30, absolute grading system will be followed. The mark range, letter grade and grade point are given in Table 6.

**Table 6 Absolute Grade: Mark Range, Letter Grade and Grade Point**

Mark Range	Letter Grade	Grade Point (g)
91 - 100	O (Outstanding)	10
81 - 90	A+ (Excellent)	9
71 - 80	A (Very Good)	8
61 - 70	B+ (Good)	7
56 - 60	B (Average)	6
50 - 55	C (Satisfactory)	5
< 50	RA (Re-appearance)	0
-	SA (Shortage of Attendance)	0
-	W (Withdrawal)	0

**8.3.3.** For NPTEL online courses, the mark range, letter grade, and grade point applicable in case of credit transfer is given in Table 7.

**Table 7 NPTEL Grade: Mark Range, Letter Grade and Grade Point**

Mark Range	Letter Grade	Grade Point (g)
90 - 100	O	10
80 - 89	A+	9
70 - 79	A	8
60 - 69	B+	7
50 - 59	B	6
40 - 49	C	5

**8.3.4. Cumulative Grade Point Average:** After the completion of the programme, the cumulative grade point average (CGPA) from the semester in which the student has joined first (first semester for regular and third semester for lateral entry students) to the final semester is calculated using the relationship:

$$CGPA = \frac{\sum g_i \times C_i}{\sum C_i}$$

where,  $g_i$  is grade point secured for  $i^{\text{th}}$  course.  $C_i$  is credit allotted for the  $i^{\text{th}}$  course.

### 8.3.5. Passing a course

A student shall be deemed to have passed a theory, laboratory, laboratory-integrated theory course and project work II with continuous assessment and end-semester examination components:

**8.3.5.1.** If he/she secures a minimum of 45% of total marks in the end-semester examination and

**8.3.5.2.** If he/she secures a minimum of 50% of total marks (continuous assessment + end-semester examination put together).

A student shall be deemed to have passed in any other course carrying only continuous assessment marks, if the total mark secured by him/her is at least 50% of total marks.

**8.3.5.3.** If a student fails to secure a pass in a theory, laboratory and laboratory-integrated theory course, the student shall register and appear only for the end-semester examination in the subsequent semester. In such cases, the continuous assessment marks obtained by the candidate in the first appearance shall be retained and considered valid for all subsequent attempts till the candidate secures a pass. However, from the third attempt (current semester's end-semester examination is considered as the first attempt) onwards if a candidate fails to obtain pass marks (continuous assessment + end-semester examination), then the candidate shall be declared to have passed the examination if he/she secure a minimum of 50% marks prescribed for the end-semester examinations alone.

**8.3.5.4.** If the course in which the student has failed is a professional elective or an open elective course, the student may be permitted to register for the same course and attend the end-semester examination or register for any other professional elective or open elective course respectively, in the subsequent semesters, and attend the classes and fulfil the attendance requirements as specified in *section 5*.

**8.3.5.5.** A student who after having earned necessary attendance is absent in the end-semester examination in any theory course (other than professional and open elective courses) has to register for the subsequent examination in that theory course when it is offered next time.

**8.3.5.6.** If the student after having earned necessary attendance is absent in the end-semester examination in any professional and open elective course, the student may be permitted to register for the same course and attend the end-semester examination or register for any other professional elective or open elective course respectively, in the subsequent semesters, and attend the classes and fulfil the attendance requirements as specified in *section 5*.

**8.3.5.6.1.** A student who after having earned necessary attendance is absent for end-semester examination or has failed in any other course such as laboratory course, mini-project work I, mini-project work II, etc. with continuous assessment and end-semester examination components will be solely assessed based on the end-semester examinations when it is offered next time.

- 8.3.5.6.2.** For any course carrying only continuous assessment marks, if the student does not meet the attendance requirements or fails to secure at least 50% of total marks, the student must register in the subsequent semester when the course is offered and attend the class and fulfil the attendance requirements and pass the examinations.
- 8.3.5.6.3.** A student who has earned necessary attendance in the course project work but does not submit the report on project work on or before the date specified by the college/department, he/she shall be deemed to have failed in the project work and will be awarded grade RA. He/she will have to register for the same at the beginning of the subsequent semester, redo and submit the project report at the end of that semester and appear for the end-semester examination, after earning the continuous assessment mark afresh.
- 8.3.5.6.4.** A student who has earned necessary attendance in the course project work but whose project report is not accepted for reasons of incompleteness or other serious deficiencies will be treated as absent and will have to register for the same at the beginning of the subsequent semester, redo and submit the project report at the end of that semester and appear for the end semester examinations, after earning the continuous assessment mark afresh.
- 8.3.5.6.5.** A student who has submitted the report on project work, but could not appear for the end-semester examination on the scheduled date, shall be deemed to have failed in the project work and awarded grade RA and will have to register for the same at the beginning of the subsequent semester, redo and submit the project report at the end of that semester and appear for the end-semester examination, after earning the continuous assessment mark afresh. The same shall be applicable also to candidates who fail to secure pass marks in the project work.
- 8.3.5.6.6.** A student who is not eligible to write the end-semester examination in any course due to lack of attendance, will be awarded grade SA and the student has to register for that course again, when offered next, attend the classes and fulfil the attendance requirements as per *section 5*. If the course, in which the student has lack of attendance, is a professional elective or an open elective, the student may register for the same or any other professional elective or open elective course respectively in the subsequent semesters.
- 8.3.5.6.7.** A student after registering for a course may withdraw his/her registration between first and second continuous assessment test for valid reasons.
- 8.3.5.6.8.** In case of credit transfer while undergoing programme in other universities/institutions, as approved by the head of the institution, out of the required six professional electives to be studied, a student has to study a minimum of three professional electives from the list of professional electives prescribed in their scheme of courses of study The remaining three professional electives can be studied either from the list of electives prescribed in the scheme of study of the department or as online courses/special courses by obtaining equivalence or by studying the required number of one-credit courses.

If a student has studied more than six professional electives totally, the six courses with the next highest grade among all remaining courses will be considered for calculation of CGPA. The grades obtained in all other remaining courses will also appear in the grade sheet.

- 8.3.5.6.9.** If a student has studied more than two open elective courses, then two open elective courses with highest grades alone will be considered for CGPA calculation. The grades obtained in other elective courses will also appear in the grade sheet.
- 8.3.5.6.10.** If a student does not clear a one- credit course it will be treated as a course withdrawn by a student; One-credit courses will be evaluated by the course instructor / department faculty member concerned, and will carry a total of 100 marks for continuous assessment; out of which 75 marks will be for final test to be scheduled by the course instructor / department faculty concerned.
- 8.3.5.6.11.** A student who is absent in the end-semester examination of a course after registering for the same will be considered to have appeared and failed in that examination and awarded grade RA.
- 8.3.5.7.** For reappearance examinations / examinations in any course under redo category, absolute grading will be followed for that course.

### **8.3.6. Scheme of Evaluation**

#### **8.3.6.1. Theory Courses with Tutorial Component (CA: 40% + ESE: 60% Total: 100 Marks)**

<b>Evaluation Component</b>	<b>Internal Assessment Marks</b>
Internal test – I (for 60 marks)	10
Internal test – II (for 60 marks)	10
Assessment tutorial - I (for 20 marks)	5
Assessment tutorial – II (for 20 marks)	5
Objective test (for 20 marks)	10
<b>Total internal assessment marks</b>	<b>40</b>

The end-semester examination is for 100 marks

For theory courses with tutorial components, a separate tutorial notebook for each course is to be maintained by the students. Assessment tutorial - I and II are of open book type, to be conducted as per schedule. During tutorial sessions, if requested the students may be guided by faculty member to solve problems.

**8.3.6.2. Theory Courses without Tutorial Component (CA: 40% + ESE: 60% Total: 100 marks)**

<b>Evaluation Component</b>	<b>Internal Assessment Marks</b>
Internal test – I (for 60 marks)	10
Internal test – II (for 60 marks)	10
Assignment – I (for 20 marks)	5
Assignment – II (for 20 marks)	5
Objective test (for 20 marks)	10
<b>Total internal assessment marks</b>	<b>40</b>

The end-semester examination is for 100 marks

Objective tests are to be conducted using LMS platform, as per schedule in the allotted halls. There will be one objective test for all the theory courses offered in the given semester, with ten questions from each course, and marks obtained course wise, will be considered for the respective courses.

**8.3.6.3. Laboratory Courses (CA: 60% + ESE: 40% Total: 100 marks)**

<b>Evaluation Component</b>	<b>Internal Assessment Marks</b>
Pre-laboratory work, observation and record work	40
Model examination	20
<b>Total internal assessment marks</b>	<b>60</b>
<b>End semester examination</b> Laboratory examination	30
Viva-voce	10
<b>Total end semester examination marks</b>	<b>40</b>

**8.3.6.4. Internship / Community Project (CA: 60% + ESE: 40% Total Marks: 100)**

<b>Evaluation Component</b>	<b>Internal Assessment Marks</b>
Presentation - I	
Guide	10
Committee	10
Presentation - II	
Guide	10
Committee	10
Report	
Guide	20
<b>Total internal assessment marks</b>	<b>60</b>
<b>End semester examination</b>	
Presentation	10
Viva-voce	10
Report evaluation	20
<b>Total end semester examination marks</b>	<b>40</b>

For the internship, the end semester examination is to be conducted by the industry personnel where the student completed the internship, in the presence of the tutor. The industry personnel may be present online. For the community project, the end semester examination is to be conducted by an external examiner recommended by the head of the department.

**8.3.6.5. Mini-project - I and Mini-project - II (CA: 60% + ESE: 40% Total Marks: 100)**

<b>Evaluation Component</b>	<b>Internal Assessment Marks</b>
Presentation - I	
Guide	10
Committee	10
Presentation - II	
Guide	10
Committee	10
Report	
Guide	20
<b>Total internal assessment marks</b>	<b>60</b>
<b>End semester examination</b>	
Presentation	10
Viva-voce	10
Report evaluation	20
<b>Total end semester examination marks</b>	<b>40</b>

End semester examination is to be conducted by an industry personnel / external subject expert recommended by the head of the department, in the presence of the guide.

**8.3.6.6. Project work I and Project work II (CA: 60% + ESE: 40% Total Marks: 100)**

<b>Evaluation Component</b>	<b>Internal Assessment Marks</b>
Presentation - I	
Guide	10
Committee	10
Presentation - II	
Guide	10
Committee	10
Thesis Evaluation	20
<b>Total internal assessment marks</b>	<b>60</b>
<b>End semester examination</b>	
Presentation	10
Viva-voce	10
Report evaluation	10
Report evaluation by guide	10
<b>Total End semester examination marks</b>	<b>40</b>

End semester examination is to be conducted by an industry personnel / external faculty member recommended by the head of the department, in the presence of the guide.

**8.3.6.7. One-Credit Course (CA: 40% + Final Examination (Internal): 60% Total: 100 Marks)**

<b>Evaluation Component</b>	<b>Internal Assessment Marks</b>
Internal test – I (20 MCQ for 45 minutes)	20
Internal test – II (20 MCQ for 45 minutes)	20
<b>Total internal assessment marks</b>	<b>40</b>
<b>Final examination (Internal) (2 hours)</b>	<b>60</b>

**8.3.6.8. Mandatory Courses I and II (CA: 100% Total Marks: 100)**

<b>Evaluation Component</b>	<b>Internal Assessment Marks</b>
Internal test - I	50
Internal test - II	50
<b>Total internal assessment marks</b>	<b>100</b>

**8.3.6.9. AICTE Activity Point Programme Courses (CA: 100%)**

<b>Evaluation Component</b>	<b>Internal Assessment Marks</b>
Report - I	50
Report - II	50
<b>Total internal assessment marks</b>	<b>100</b>

Every student is required to prepare and submit a report with documentary proof of activities for each assessment component.

**8.3.6.10. Workplace communication skills / Foundations of problem solving / Problem solving / Aptitude skills / Enhancing problem solving ability with code / Enhancing arithmetic problem solving (CA: 100% Total Marks: 100)**

<b>Evaluation Component</b>	<b>Internal Assessment Marks</b>
Internal test - I	50
Internal test - II	50
<b>Total internal assessment marks</b>	<b>100</b>

**8.3.6.11. Online courses (ESE: 100%)**

For online courses, grading will be done as specified in *section 8.3.3*.

**8.3.6.12. Self-directed learning courses**

For self-directed learning courses, grading will be done as specified in *section 4.5*.

**9. CONDUCT OF ACADEMIC AUDIT BY THE DEPARTMENT**

All faculty members shall strive to improve the performance of students by assessing their performance in their respective internal assessment components and provide appropriate individual feedback to students for improvement. The head of the department shall nominate two senior faculty members as department's internal academic auditors. The internal academic auditors shall assess the academic activities of the department twice in a semester at appropriate intervals and submit the report to the head of the institution through the head of the department.

Similarly, the performance of the students in their end-semester examinations shall be evaluated after the publication of the result, in the department meeting, and appropriate feedback shall be given to the students by their respective faculty mentors.

The head of the department shall arrange to conduct another academic audit once every year for all the courses conducted through an external expert, recommended by the head of the department and approved by the head of the institution.

All academic audits shall cover verification of all the academic records pertaining to the regulation, attendance and assessment, attainment of course outcomes (and programme outcomes where applicable), and assessment of the overall teaching-learning process. Appropriate action shall be taken based on the audit reports, for improving the academic process. Academic documents should be available with the faculty member / department for 5 years.

## **10. QUALIFICATION FOR AWARD OF DEGREE**

A student shall be declared to have qualified for the award of B.E./B.Tech. degree provided:

- 10.1.** The student has successfully completed the course requirements and has passed all the prescribed courses of study of the respective programme.
- 10.2.** No disciplinary action is pending against the student

## **11. CLASSIFICATION OF DEGREE**

Classification of a student while awarding the degree will not be affected if the student has to redo courses which are mandatory in nature (i.e. having no credit but whose completion is compulsory for the award of degree).

### **11.1. First class with distinction**

A student who satisfies the following conditions shall be declared to have passed the examination in first class with distinction:

- 11.1.1.** Should have passed the end-semester examination in all the courses of all the eight semesters (six semesters in the case of lateral entry) in his/her first appearance within five years (four years in the case of lateral entry), which includes authorised break of study of one year. The authorised withdrawal from examination will not be considered as an appearance.
- 11.1.2.** Should have secured a CGPA of not less than 8.50.
- 11.1.3.** One year authorised break of study (if availed) is included in the five years (four years in the case of lateral entry) for award of first class with distinction.

**11.1.4.** Should not have been prevented from writing end-semester examination due to lack of attendance in any of the courses.

### **11.2. First class**

A student who satisfies the following conditions shall be declared to have passed the examination in first class:

**11.2.1.** Should have passed the examination in all the courses of all eight semesters (six semesters in the case of lateral entry) within five years (four years in the case of lateral entry).

**11.2.2.** One year authorised break of study (if availed) or prevention from writing the end-semester examination due to lack of attendance (if applicable) is included in the duration of five years (four years in the case of lateral entry) for award of first class.

**11.2.3.** Should have secured a CGPA of not less than 6.5.

### **11.3. Second class**

All other students who qualify for the award of the degree shall be declared to have passed the examination in second class.

### **11.4. B.E. /B.Tech. (honours) and B.E. /B.Tech. minor degree in another discipline**

**11.4.1.** For B.E./B.Tech. with honours degree in the same discipline, students should have earned additionally a minimum of 18 credits from one or more verticals of the same programme, from 5<sup>th</sup> semester to 7<sup>th</sup> semester. He/she should have passed all the courses in the first attempt and should have earned a minimum CGPA of 8.50 for the award of degree in first class with distinction, or a minimum CGPA 7.50 for the award of degree in first class.

**11.4.2.** If a student decides not to opt for the honours degree, after completing a certain number of additional courses, such additional courses studied shall be considered instead of the professional elective courses which are part of the curriculum. If the student has studied more of such courses than the number of professional elective courses required as per the curriculum, the courses with highest grades shall be considered for the calculation of CGPA. Remaining courses shall be printed in the grade sheet; however, they will not be considered for calculation of CGPA. If the student has failed in the additional courses or faced shortage of attendance, they will not be printed in the grade sheet and will not be considered for CGPA calculation and classification of degree.

**11.4.3.** For B.E./B.Tech. with minor degree, students should have earned additionally a minimum of 18 credits from one or more verticals of programme offered by the department other than the parent department, from 5<sup>th</sup> semester to 7<sup>th</sup> semester. He/she should have passed all the courses in the first attempt and should have earned a minimum CGPA of 8.50 for the award of degree in first class with distinction, or a minimum CGPA 7.50 for the award of degree in first class.

**11.4.4.** If a student decides not to opt for the minor degree, after completing a certain number of courses, the additional courses studied shall be considered instead of open elective courses which are part of the curriculum. If the student has studied more of such courses than the number of open electives required as per the curriculum, the courses with highest grades shall be considered for calculation of CGPA. Remaining courses shall be printed in the grade sheet; however, they will not be considered for calculation of CGPA. If the student has failed in the additional courses or faced shortage of attendance, they will not be printed in the grade sheet and will not be considered for CGPA calculation and classification of degree

## **12. WITHDRAWAL FROM EXAMINATION**

- 12.1.** A student may, for valid reasons, be granted permission to withdraw from appearing for the examination in any course or courses of only one semester if he/she does not have any history of reappearance courses at the time of request for withdrawal. Prior permission for withdrawal from semester examinations is to be obtained from the head of the institution. Only one application for withdrawal is permitted for end-semester examinations in which withdrawal is sought, for the entire duration of study whatever the reasons may be.
- 12.2.** Withdrawal application shall be valid only if the student is otherwise eligible to write the examination and if it is made prior to the commencement of the examination in that course or courses, and also recommended by the head of the department.

## **13. TEMPORARY BREAK OF STUDY FROM THE PROGRAMME**

- 13.1.** A student is permitted to go on break of study for a fixed period of one year as a single break in the entire course of study.
- 13.2.** A student is not normally permitted to temporarily break the study. However, if a student intends to temporarily discontinue the programme, the student has to apply to the head of the institution through the head of the department, not later than the last date of the first assessment period. Notwithstanding the requirement of mandatory first assessment period, applications for break of study for special cases viz., prolonged hospitalisation, accidents will be considered on the merit of the case
- 13.3.** A student is permitted to rejoin the programme at the respective semester as and when it is offered after the break, subject to the approval of Commissioner of Technical Education and Anna University, Chennai, and shall be governed by rules and regulations in force at the time of rejoining.
- 13.4.** The duration specified for passing all the courses for the purpose of classification shall be increased by the period of such break of study permitted.
- 13.5.** The total period for completion of the programme reckoned from the commencement of the semester to which the student was first admitted shall not exceed the maximum period irrespective of the period of break of study in order that he/she may be qualified for the award of the degree.

- 13.6.** If any student is detained for want of requisite attendance, progress and conduct, the period spent in that semester shall not be considered as permitted break of study and is not applicable for such cases.

#### **14. FACULTY MENTOR**

To help students in planning their courses of study and for general guidance for completing the academic programme and for possible career opportunities, the head of the student department will assign a group of students to a faculty member, who shall function as faculty mentor for the students throughout the period of study. The faculty mentor shall advise the mentees, monitor their attendance and progress, and counsel them periodically. If necessary, the faculty mentor may also discuss with or inform the respective parents about the progress of the mentees concerned, The number of students assigned to a faculty mentor will be decided by the head of the department, and the number, in general, shall not exceed 30.

- 14.1.** The responsibilities of the faculty mentor shall be:

- To help students understand the regulations of study, and their rights and responsibilities.
- To inform code of conduct to be maintained in the campus.
- To inform the students about the various facilities and activities available to enhance the student's curricular and co-curricular activities.
- To guide students in choosing courses for registration.
- To monitor the academic and general performance of the students including attendance and to counsel them accordingly.
- To collect and maintain the academic and co-curricular records of the students.
- To facilitate and collect students' feedback about the course and course instructor, and the programme's exit survey.
- To provide all the details of the academic programme including feedback, training, scholarship, placement and co-curricular and extra-curricular activities of the students to the department.

#### **15. COURSE COMMITTEES**

A theory course handled by more than one faculty member shall have a common course committee, composed of all the faculty members teaching that course. One of the senior faculty members shall be nominated as course coordinator by the head of the institution. The committee shall be constituted within 15 days from the commencement of the semester. The first meeting of the committee shall be held within fifteen days from the date of formation of the committee. The lesson plan, books and references to be followed shall be decided in the first meeting. Two or three subsequent meetings in a semester may

be held at suitable intervals. In addition, the committee shall meet to ensure uniform evaluation of continuous assessments after arriving at a common scheme of evaluation for the assessments. Wherever feasible, the common course committee shall prepare a common question paper for the continuous assessment tests also. The question paper for the end-semester examination will be a common question paper.

## **16. CLASS COMMITTEES**

The class committee shall comprise all the faculty members handling courses of a particular semester for the class, and a minimum of 6 students (with the proper mix of girls and boys, as applicable) from the class. One of the faculty mentors of the class, nominated by the head of the department as class advisor shall coordinate the class committee activities. The committee shall be constituted by the head of the department within 10 days from the commencement of semester. The class committee meeting will be chaired by the head of the department. The class advisor shall prepare the minutes of the meeting.

**16.1.** The functions of the class committee are as follows:

- The first meeting of the class committee shall be held within 10 days from the date of formation of the committee.
- Two or three subsequent meetings in a semester may be held at suitable intervals.
- During these meetings, the student members shall meaningfully interact and express their opinions and suggestions on behalf of all the students to improve the effectiveness of the teaching-learning process.

## 17. COURSES OF STUDY AND SCHEME OF ASSESSMENT

### B.E. CIVIL ENGINEERING (Minimum No. of credits to be earned: 168)

S. No.	Course Code	Course Title	Hours / Week			Credits	Maximum Marks			CAT
			Lecture	Tutorial	Practical		CA	ESE	Total	
<b>SEMESTER I</b>										
<b>THEORY</b>										
1	25MA101	Calculus and its Applications	3	1	0	4	40	60	100	BS
2	25PH101	Physics for Civil Engineering	3	0	0	3	40	60	100	BS
3	25CY101	Chemistry of Building Materials I	3	0	0	3	40	60	100	BS
4	25CE101	Engineering Geology	3	0	0	3	40	60	100	ES
5	25HS101	English Language Proficiency	3	1	0	4	40	60	100	HS
6	25HS102	தமிழர் மரபு / Heritage of Tamils	1	0	0	1	40	60	100	HS
<b>PRACTICALS</b>										
7	25BS111	Basic Sciences Laboratory I	0	0	4	2	60	40	100	BS
8	25CE111	Engineering Drawing	0	0	4	2	60	40	100	ES
9	25GE111	Design Thinking for Innovation	0	0	2	1	100	0	100	ES
<b>MANDATORY COURSES</b>										
10	25GEM01	Induction Programme**	-	-	-	Grade	-	-	-	MC
<b>Total 28 periods</b>			<b>16</b>	<b>2</b>	<b>10</b>	<b>23</b>	<b>460</b>	<b>440</b>	<b>900</b>	

S. No.	Course Code	Course Title	Hours / Week			Credits	Maximum Marks			CAT
			Lecture	Tutorial	Practical		CA	ESE	Total	
<b>SEMESTER II</b>										
<b>THEORY</b>										
1	25MA201	Complex Variables and Transforms	3	1	0	4	40	60	100	BS
2	25CE201	Engineering Statics and Dynamics	3	1	0	4	40	60	100	ES
3	25PH201	Applied Physics	3	0	0	3	40	60	100	BS
4	25CY201	Chemistry of Building Materials II	3	0	0	3	40	60	100	BS
5	25HS201	தமிழரும் தொழில்நுட்பமும் / Tamils and Technology	1	0	0	1	40	60	100	HS
<b>PRACTICALS</b>										
6	25HS21_	Language Elective	0	0	4	2	60	40	100	HS
7	25BS211	Basic Sciences Laboratory II	0	0	4	2	60	40	100	BS
8	25CE211	Engineering Practices Laboratory	0	0	2	1	60	40	100	ES
9	25CE212	Problem Solving using Python Programming Laboratory	0	0	4	2	60	40	100	ES
10	25EEC01	Workplace Communication Skills	0	0	2	Grade	100	0	100	EEC
<b>MANDATORY COURSES</b>										
11	25GEM02	Activity Point Programme I*	-	-	-	Grade	-	-	-	MC
<b>Total 31 periods</b>			<b>13</b>	<b>2</b>	<b>16</b>	<b>22</b>	<b>540</b>	<b>460</b>	<b>1000</b>	

\*\*As per AICTE Norms

\* As per AICTE Norms; Total: 60 hrs.; Grade: Non-credit Course

CAT – Category; BS – Basic Science; HS – Humanities and Social Sciences; ES – Engineering Sciences; PC – Professional Core; PE – Professional Elective; OE – Open Elective; EEC – Employability Enhancement Course; MC – Mandatory Course; CA – Continuous Assessment; ESE – End Semester Examination

S. No.	Course Code	Course Title	Hours / Week			Credits	Maximum Marks			CAT
			Lecture	Tutorial	Practical		CA	ESE	Total	
<b>SEMESTER III</b>										
<b>THEORY</b>										
1	25MA304	Matrix Theory and Numerical Methods	3	1	0	4	40	60	100	BS
2	25CE301	Mechanics of Solids I	3	1	0	4	40	60	100	ES
3	25CE302	Construction Materials and Practices	3	0	0	3	40	60	100	PC
4	25CE303	Surveying	3	1	0	4	40	60	100	PC
5	25HS301	Project and Finance Management	3	0	0	3	40	60	100	HS
<b>PRACTICALS</b>										
6	25CE311	Strength of Materials Laboratory	0	0	2	1	60	40	100	ES
7	25CE312	Survey Practice	0	0	4	2	60	40	100	PC
8	25EEC02	Foundations of Problem Solving	0	0	2	1	100	0	100	EEC
<b>MANDATORY COURSES</b>										
9	25MC0__	Mandatory Course I	2	0	0	Grade	100	0	100	MC
10	25GEM03	Activity Point Programme II*	-	-	-	Grade	-	-	-	MC
<b>Total 28 periods</b>			<b>17</b>	<b>3</b>	<b>8</b>	<b>22</b>	<b>520</b>	<b>380</b>	<b>900</b>	

S. No.	Course Code	Course Title	Hours / Week			Credits	Maximum Marks			CAT
			Lecture	Tutorial	Practical		CA	ESE	Total	
<b>SEMESTER IV</b>										
<b>THEORY</b>										
1	25CE401	Mechanics of Solids II	3	1	0	4	40	60	100	ES
2	25CE402	Hydraulic Engineering	3	1	0	4	40	60	100	ES
3	25CE403	Basic Structural Steel Design	3	1	0	4	40	60	100	PC
4	25CE404	Highway and Railway Engineering	3	0	0	3	40	60	100	PC
5	25__O__	Open Elective I	3	0	0	3	40	60	100	OE
<b>PRACTICALS</b>										
6	25CE411	Concrete Technology and Highway Laboratory	0	0	4	2	60	40	100	PC
7	25CEE01	Mini Project I	0	0	2	1	100	0	100	EEC
8	25EEC03	Problem Solving	0	0	2	1	100	0	100	EEC
<b>MANDATORY COURSES</b>										
9	25MC0__	Mandatory Course II	2	0	0	Grade	100	0	100	MC
10	25GEM04	Activity Point Programme III*	-	-	-	Grade	-	-	-	MC
<b>Total 28 periods</b>			<b>17</b>	<b>3</b>	<b>8</b>	<b>22</b>	<b>460</b>	<b>340</b>	<b>800</b>	

\* As per AICTE Norms; Total: 60 hrs.; Grade: Non-credit Course

CAT – Category; BS – Basic Science; HS – Humanities and Social Sciences; ES – Engineering Sciences; PC – Professional Core; PE – Professional Elective; OE – Open Elective; EEC – Employability Enhancement Course; MC – Mandatory Course; CA – Continuous Assessment; ESE – End Semester Examination

S. No.	Course Code	Course Title	Hours / Week			Credits	Maximum Marks			CAT
			Lecture	Tutorial	Practical		CA	ESE	Total	
<b>SEMESTER V</b>										
<b>THEORY</b>										
1	25CE501	Structural Analysis I	3	1	0	4	40	60	100	PC
2	25CE502	Design of RC Elements	3	1	0	4	40	60	100	PC
3	25CE503	Mechanics of Soil	3	1	0	4	40	60	100	PC
4	25CE504	Water Supply Engineering	3	0	0	3	40	60	100	PC
5	25 __O__	Open Elective II	3	0	0	3	40	60	100	OE
<b>PRACTICALS</b>										
6	25CE511	Soil Mechanics Laboratory	0	0	2	1	60	40	100	PC
7	25CE512	Hydraulics and Hydraulic Machinery Laboratory	0	0	2	1	60	40	100	ES
8	25CEE02/ 25CEE03	Internship I/Community Project	0	0	0	1	100	0	100	EEC
9	25EEC04	Aptitude Skills	0	0	2	1	100	0	100	EEC
<b>MANDATORY COURSES</b>										
10	25GEM05	Activity Point Programme IV*	-	-	-	Grade	-	-	-	MC
<b>Total 24 periods</b>			<b>15</b>	<b>3</b>	<b>6</b>	<b>22</b>	<b>520</b>	<b>380</b>	<b>900</b>	

S. No.	Course Code	Course Title	Hours / Week			Credits	Maximum Marks			CAT
			Lecture	Tutorial	Practical		CA	ESE	Total	
<b>SEMESTER VI</b>										
<b>THEORY</b>										
1	25CE601	Structural Analysis II	3	1	0	4	40	60	100	PC
2	25CE602	Construction Project Management	3	0	0	3	40	60	100	PC
3	25CE603	Waste Water Engineering	3	1	0	4	40	60	100	PC
4	25CE604	Foundation Engineering	3	0	0	3	40	60	100	PC
5	25CE605	Design of Steel Structures	3	1	0	4	40	60	100	PC
<b>PRACTICALS</b>										
6	25CE611	Environmental Engineering Laboratory	0	0	4	2	60	40	100	PC
7	25CE612	Building Planning and Drafting Laboratory	0	0	2	1	60	40	100	PC
8	25CEE04	Mini Project II	0	0	2	1	100	0	100	EEC
9	25EEC05	Enhancing Problem Solving Ability with Code	0	0	2	1	100	0	100	EEC
<b>MANDATORY COURSES</b>										
10	25GEM06	Activity Point Programme V*	-	-	-	Grade	-	-	-	MC
<b>Total 29 periods</b>			<b>15</b>	<b>3</b>	<b>10</b>	<b>23</b>	<b>520</b>	<b>380</b>	<b>900</b>	

\* As per AICTE Norms; Total: 60 hrs.; Grade: Non-credit Course

CAT – Category; BS – Basic Science; HS – Humanities and Social Sciences; ES – Engineering Sciences; PC – Professional Core; PE – Professional Elective; OE – Open Elective; EEC – Employability Enhancement Course; MC – Mandatory Course; CA – Continuous Assessment; ESE – End Semester Examination

S. No.	Course Code	Course Title	Hours / Week			Credits	Maximum Marks			CAT
			Lecture	Tutorial	Practical		CA	ESE	Total	
<b>SEMESTER VII</b>										
<b>THEORY</b>										
1	25CE701	Estimation and Costing	3	1	0	4	40	60	100	PC
2	25CEP__	Professional Elective I	3	0	0	3	40	60	100	PE
3	25CEP__	Professional Elective II	3	0	0	3	40	60	100	PE
4	25CEP__	Professional Elective III	3	0	0	3	40	60	100	PE
5	25CEP__	Professional Elective IV	3	0	0	3	40	60	100	PE
<b>PRACTICALS</b>										
6	25CE711	Design and Detailing of Structures	1	0	4	3	50	50	100	PC
7	25CE712	Computer Analysis and Design Laboratory	0	0	4	2	60	40	100	PC
8	25CEE05	Project Work I	0	0	4	2	100	0	100	EEC
9	25CEE06	Internship II	0	0	0	1	100	0	100	EEC
<b>Total 29 periods</b>			<b>16</b>	<b>1</b>	<b>12</b>	<b>24</b>	<b>510</b>	<b>390</b>	<b>900</b>	

S. No.	Course Code	Course Title	Hours / Week			Credits	Maximum Marks			CAT
			Lecture	Tutorial	Practical		CA	FE	Total	
<b>SEMESTER VIII</b>										
<b>THEORY</b>										
1	25CEP__	Professional Elective V	3	0	0	3	40	60	100	PE
2	25CEP__	Professional Elective VI	3	0	0	3	40	60	100	PE
<b>PRACTICALS</b>										
3	25CEE07	Project Work II	0	0	8	4	60	40	100	EEC
<b>Total 14 periods</b>			<b>6</b>	<b>0</b>	<b>8</b>	<b>10</b>	<b>140</b>	<b>160</b>	<b>300</b>	

CAT – Category; BS – Basic Science; HS – Humanities and Social Sciences; ES – Engineering Sciences; PC – Professional Core; PE – Professional Elective; OE – Open Elective; EEC – Employability Enhancement Course; MC – Mandatory Course; CA – Continuous Assessment; ESE – End Semester Examination

#### SUMMARY OF CREDIT DISTRIBUTION

<b>BE CIVIL ENGINEERING</b>										
S. No.	Course Category	Credits Per Semester								Total Credits
		I	II	III	IV	V	VI	VII	VIII	
1	HS	5	3	3	0	0	0	0	0	11
2	BS	12	12	4	0	0	0	0	0	28
3	ES	6	7	5	8	1	0	0	0	27
4	PC	0	0	9	9	16	21	9	0	64
5	PE	0	0	0	0	0	0	12	6	18
6	OE	0	0	0	3	3	0	0	0	6
7	EEC	0	0	1	2	2	2	3	4	14
8	MC	-	-	-	-	-	-	-	-	-
<b>TOTAL</b>		<b>23</b>	<b>22</b>	<b>22</b>	<b>22</b>	<b>22</b>	<b>23</b>	<b>24</b>	<b>10</b>	<b>168</b>

CAT – Category; BS – Basic Science; HS – Humanities and Social Sciences; ES – Engineering Sciences; PC – Professional Core; PE – Professional Elective; OE – Open Elective; EEC – Employability Enhancement Course; MC – Mandatory Course

## LIST OF PROFESSIONAL ELECTIVE COURSES: VERTICALS

S. No.	Vertical – I (Structural Engineering and Construction Practices)	Vertical – II (Environmental and Water Resources Engineering)	Vertical – III (Geo-informatics and Geo- technical Engineering)	Vertical – IV (Diversified)
1	25CEP01 Prefabricated Structures	25CEP09 Air Pollution and Control Engineering	25CEP17 Ground Improvement Techniques	25CEP25 Airport Docks and Harbour Engineering
2	25CEP02 Finite Element Analysis	25CEP10 Environmental Impact Assessment	25CEP18 Total Station and GPS Surveying	25CEP26 Housing Planning and Management
3	25CEP03 Structural Dynamics and Earthquake Resistant Design	25CEP11 Industrial Waste Management	25CEP19 Geographic Information Systems	25CEP27 Traffic Engineering, Safety and Management
4	25CEP04 Prestressed Concrete Structures	25CEP12 Hydrology and Water Resources Engineering	25CEP20 Remote Sensing Techniques and Applications	25CEP28 Smart City Planning and Development
5	25CEP05 Repair and Rehabilitation of Structures	25CEP13 Solid Waste Management	25CEP21 Geosynthetics in Civil Engineering	25CEP29 Design of Energy Efficient Buildings
6	25CEP06 Sustainable and Lean Construction	25CEP14 Irrigation Engineering	25CEP22 Pavement Engineering	25CEP30 Digital Construction Techniques
7	25CEP07 Advanced Reinforced Concrete Design	25CEP15 Watershed Conservation and Management	25CEP23 Disaster Management and Mitigation	25CEP31 Architecture for Civil Engineering
8	25CEP08 Advanced Concrete Technology	25CEP16 Groundwater Engineering	25CEP24 Cartography	25CEP32 Environment, Health and Safety

## LIST OF PROFESSIONAL ELECTIVE COURSES FOR MINOR DEGREE PROGRAMME

S. No.	Course Code	Course Title
1	25CEM01	Construction Engineering Practices
2	25CEM02	Estimation, Costing and Valuation
3	25CEM03	Geotechnical Engineering
4	25CEM04	Surveying Techniques
5	25CEM05	Transportation Engineering
6	25CEM06	Water and Wastewater Engineering
7	25CEM07	Pollution Control and Management
8	25CEM08	Waste Management for Circular Economy
9	25CEM09	Structural Analysis and Design

**25MA101 CALCULUS AND ITS APPLICATIONS**  
(Common to CIVIL, CSE, EEE, ECE, ICE, MECH, AI&DS and EE-VLSI)

**3 1 0 4**

**DIFFERENTIAL CALCULUS:** Functions of two variables, limits and continuity, partial derivatives, chain rule, extreme values and saddle points, Lagrange multipliers, Taylor's formula for two variables. **(9+3)**

**INTEGRAL CALCULUS:** Double and iterated integrals over rectangles, double integrals over general regions, Fubini's theorem, area and volume by double integration, reversing the order of integration, double integrals in polar form. **(9+3)**

**FIRST ORDER ORDINARY DIFFERENTIAL EQUATIONS:** Basic concepts, separable differential equations, exact differential equations, integrating factors, linear differential equations, modeling - mixing problems, Newton's law of cooling, decay and growth problems. **(9+3)**

**SECOND ORDER LINEAR DIFFERENTIAL EQUATIONS:** Homogeneous linear equations of second order, homogeneous linear ODEs with constant coefficients, Euler–Cauchy equations, solution by variation of parameters, free oscillations mass spring systems, electric circuits. **(9+3)**

**VECTOR CALCULUS:** Gradient and directional derivative of a scalar field, divergence and curl of a vector field. Integration in vector field – line integrals, path independence of line integrals, Green's theorem in the plane, divergence theorem of Gauss and Stokes' theorem. **(9+3)**

**Total L: 45 + T: 15 = 60 periods**

**TEXT BOOKS**

1. J. Hass, C. Heil, and D.W. Maurice '*Thomas' Calculus*'. Pearson Education, New Delhi, 2018.
2. Erwin Kreyszig '*Advanced Engineering Mathematics*'. Wiley India, New Delhi, 2018.

**REFERENCES**

1. H. Anton, I. Bivens, and S. Davis '*Calculus*'. John Wiley and Sons, USA, 2016.
2. C. R. Wylie and L.C. Barrett, '*Advanced Engineering Mathematics*'. Tata McGraw-Hill, New Delhi, 2019.
3. D.G. Michael, '*Foundations of Applied Mathematics*'. Dover Publications, New York, 2013.
4. Gilbert Strang '*Calculus*' Wellesley Cambridge Press, USA, 2017.

**COURSE OUTCOMES**

At the end of the course, students will be able to:		Bloom's Level
CO1	Explain the concepts related to Calculus, Differential Equations and Vector Calculus.	K2
CO2	Apply the techniques of Calculus, Differential Equations and Vector Calculus to solve engineering problems.	K3
CO3	Analyze the solutions of engineering problems employing Calculus, Differential Equations and Vector Calculus.	K4
CO4	Use modern tools to solve engineering problems with the help of Calculus, Differential Equations and Vector Calculus.	-

**25PH101 PHYSICS FOR CIVIL ENGINEERING****3 0 0 3**

**PROPAGATION OF ELASTIC WAVES:** Simple harmonic motion. Velocity damping. Damping coefficient. Differential equation of SHM. Velocity and acceleration. Restoring force. Vibration of a spring and mass system. Frequency response, phase response and resonance. Definition of a plane progressive wave. Attenuation of waves. Differential equation of a plane progressive wave. Phase velocity. Phase and phase difference. Solution of the differential equation of a plane progressive wave. (9)

**TESTING OF MATERIALS BY ELASTIC WAVE PROPAGATION:** Impact echo method-determination of thickness of concrete and flaw detection. S, P and R waves and their dependence on elastic moduli. Ultrasonic flaw detection-Frequency and power. Pulse echo method of flaw detection. Single and phase-array transducers- waveform synthesis and scanning-A, B, C and S Scan. (9)

**LASERS:** Properties of laser radiation and their significance-wavelength, power, monochromaticity, coherence. Types of lasers-working media and their radiation characteristics-Power, wavelength and operational modes of He-Ne, Nd-YAG, Carbon-dioxide, diode lasers. Physical principles of Laser beam delivery systems. Applications-Ranging and survey, cutting and welding. (9)

**FLUID PROPERTIES & DIMENSIONAL ANALYSIS:** Dimensions and units - Fluid properties - Types of fluids - Hydrostatic law; Pascal's law - Atmospheric, Absolute, Gauge and Vacuum pressures - Measurement of pressure by various types of manometers and mechanical gauges. Methods of Dimensional Analysis - Rayleigh's method - Buckingham's theorem. (9)

**THERMAL PROPERTIES:** Specific heat capacity, thermal capacity. Coefficient of linear thermal expansion. Methods of measurement of thermal expansion. Thermal stresses in composite structures due to non-homogeneous thermal expansion. Applications -The bimetallic strip. Expansion gaps and rollers in engineering structures. (9)

**Total L: 45 periods****TEXT BOOKS**

1. M.N. Avadhanulu and P.G. Kshirsagar, A Text Book of 'Engineering Physics, 9<sup>th</sup> Revised Edition'. S. Chand and Company, New Delhi, 2018.
2. Ronald L Panton, 'Incompressible Flow, 5<sup>th</sup> Edition'. Wiley, 2024.
3. R.K. Bansal, A Textbook of 'Fluid Mechanics and Hydraulic Machines, 11<sup>th</sup> Edition'. Laxmi Publications, New Delhi, 2023.

**REFERENCES**

1. F.S. Crawford, Jr. 'Waves – Berkeley Physics Course, Vol. 3'. Tata McCraw-Hill 2008.
2. James F. Shackelford 'Introduction to Material Science for Engineers, 9<sup>th</sup> edition'. Prentice Hall 2019.
3. Raymond A. Serway and John W. Jewett 'Physics for Scientists and Engineers, 10<sup>th</sup> Edition'. Cengage Learning. 2019.

**COURSE OUTCOMES**

At the end of the course, students will be able to:		<b>Bloom's Level</b>
<b>CO1</b>	Explain the basic principles of elastic wave propagation, laser properties, fluid dynamics, and thermal expansion in engineering systems.	<b>K2</b>
<b>CO2</b>	Apply mathematical and physical principles to calculate wave velocity, pressure, and thermal expansion, and interpret measurement techniques in material testing and laser-based engineering applications.	<b>K3</b>
<b>CO3</b>	Examine the effects of elastic wave propagation, fluid pressure, thermal expansion and laser-material interactions on material integrity using experimental data and analytical techniques.	<b>K4</b>
<b>CO4</b>	Prepare a report or presentation on the practical applications of wave testing, laser systems, thermal stress analysis, and fluid pressure measurement techniques in engineering materials	-

**25CY101 CHEMISTRY OF BUILDING MATERIALS I****3 0 0 3**

**MATERIALS FOR MAKING CONCRETE – I:** Water - Hardness, Determination by EDTA method – Ion Exchange method of water softening. Water quality standards for construction. CEMENT: Chemical composition, manufacturing of Portland cement – dry and wet processes – reactions in rotary kiln. Setting and hardening of cement - hydration of cement-microstructure of cement paste. Types of Portland cements – rapid hardening cement, low heat cement, sulphate resisting cement, Super sulphated Portland cement, Pozzolana cement, Portland Slag cement. (9)

**MATERIALS FOR MAKING CONCRETE – II: AGGREGATES -** Classification and characteristics of aggregates – Strength, Stiffness, bond strength, specific gravity, bulk density, voids and porosity. Fine aggregate, Coarse aggregate, Cinder aggregate, alkali aggregate reaction – mechanism and factors affecting and control measures on alkali-aggregate reaction. Admixtures and Mineral Additives. (9)

**POLYMERS:** Classification, degree of polymerization, weight and number average molecular weights (Definition only). Polymerization reactions, step and chain mechanism. Compounding of plastics, polymer processing by injection, extrusion, compression and blow moulding techniques. Polymers in civil engineering- Reinforced Plastics – Glass Fibre Reinforced plastics, coating, polymer concretes. Thermal insulating foams. Plastic products – Floor tiles, floor finishes, PVA floor finish, Roofing – PVC and Glass fibre reinforced plastic corrugated roofing sheets. (9)

**CORROSION AND ITS PREVENTION:** Electrode potential – emf and galvanic series, corrosion rate expression, E/ pH diagram of iron, electrode polarization – passivation, polarization curves. Oxidation – mechanism, Pilling – Bedworth rule. Galvanic corrosion, differential aeration corrosion, atmospheric corrosion, pitting corrosion, waterline corrosion

and soil corrosion. Factors influencing corrosion, corrosion of rebars in concrete. Corrosion control -cathodic protection, anodic protection, selection of materials and proper designing. (9)

**PAINTS AND VARNISHES:** PAINTS – Paints – constituents and their functions, types – Water paint, Aluminum paint, Bituminous paint, Cellulose paint, Cement paint, Graphite paint, Silicate paint. Defects in Painting. Varnishes – Characteristics and ingredients of a Varnish, types of varnish – Oil varnish, turpentine varnish, Sprit varnish. Coatings –Electroplating of metals – Cu, Ni, Cr and applications. (9)

**Total L: 45 periods**

**TEXT BOOKS:**

1. P.C. Jain and M. Jain, '*Engineering Chemistry*'. Dhanpat Rai Publishing Company, New Delhi., 17<sup>th</sup> Edition, 2019.
2. U.K. Shrivastava, '*Building Materials Technology*'. Galgotia Publications Pvt Ltd, New Delhi., 1st Ed, 2012.

**REFERENCES:**

1. N. Jackson and R.K. Dhir '*Civil Engineering Materials*'. Brijbasi Art press Ltd., Noida, 2007.
2. S Shan, '*Civil Engineering Materials*'. Pearson Education. Inc – Prentice Hall., New Delhi, 2012.
3. S.K. Duggal, '*Building Materials*'. New Age International Publishers., New Delhi, 5<sup>th</sup> Edition, 2019.

**COURSE OUTCOMES**

At the end of the course, students will be able to:		Bloom's Level
CO1	Understand the basic properties of materials for making concrete.	K2
CO2	Apply the suitable materials for developing civil engineering products.	K3
CO3	Analyze the strength of materials for building construction.	K4

**25CE101 ENGINEERING GEOLOGY**

**3 0 0 3**

**GENERAL GEOLOGY:** Geology in Civil Engineering– Branches of geology–Earth Structures and composition– Elementary knowledge on continental drift and plate tectonics– Geological Time scale--Earth processes Weathering Types of soils – Work of rivers and wind and their engineering importance. Engineering Seismology –causes of earthquakes; seismic waves; magnitude, intensity and energy release, seismic zones of India-Earthquake belts in world. (9)

**MINERALOGY:** Elementary knowledge on symmetry elements of important crystallographic systems – physical properties of minerals – study of the following rock forming minerals – Quartz family. Feldspar family, Augite, Hornblende, Biotite, Muscovite, Calcite, Gypsum, Garnet – Identification of minerals - Properties, behaviour and engineering significance of clay minerals – Fundamentals of process of formation of ore minerals – Coal and petroleum – Their origin and occurrence in India. (9)

**PETROLOGY:** Classification of rocks – distinction between igneous, sedimentary and metamorphic rocks. Description, occurrence, engineering properties and distribution of following rocks. Igneous rocks – Granite, Syenite, Diorite, Gabbro, Pegmatite, Dolerite and Basalt. Sedimentary rocks - Sandstone, Limestone, Shale, Conglomerate and Breccia. Metamorphic rocks - Quartzite, Marble, Slate, Charnockite, Phyllite, Gneiss and Schist– Identification of rocks and commercial granites and marbles - Testing on rock as a construction material. (9)

**STRUCTURAL GEOLOGY AND GEOPHYSICAL METHOD:** Attitude of beds– Outcrops-Introduction to Geological maps– Study of structures – Folds, faults and joints – Their bearing on engineering construction. Seismic and Electrical methods for Civil Engineering investigations – Clinometer - Brunton Compass - Geology of India-Types and occurrence of rock formation. (9)

**GEOLOGICAL INVESTIGATIONS IN CIVIL ENGINEERING:** Remote sensing techniques - Applications in Civil Engineering - Groundwater-occurrence–investigation, quality, recharge – importance in civil engineering. Geological conditions necessary for construction of Dams, Tunnels, Buildings, Road Cuttings-Landslides–Causes and prevention. Sea erosion and coastal protection. (9)

**TOTAL L: 45 periods**

**TEXT BOOKS:**

1. Parbin Singh ‘*Engineering and General Geology*’. Katson Books, New Delhi, 2013.
2. Tony Waltham ‘*Foundations of Engineering Geology*’. Spon Press, New York, 2009.

**REFERENCES:**

1. F.G.H. Blyth and M.H. de Freitas ‘*A Geology for Engineers*’. Edward Arnold, London, 2017.
2. Gokhale ‘*Principles of Engineering Geology*’. BS Publications, Hyderabad, 2019.
3. F.G. Bell ‘*Fundamentals of Engineering Geology*’ BS Publications, Hyderabad, 2020.

**COURSE OUTCOMES**

At the end of the course, students will be able to:		Bloom’s Level
CO1	Elaborate the geological phenomenon in the earth and differentiate the various types of minerals, rocks and their sub groups based on the physiochemical properties.	K2
CO2	Interpret and Select the appropriate geophysical methods and modern techniques for surface and sub-surface explorations for mapping and construction of various civil engineering projects.	K3
CO3	Demonstrate the necessary geological conditions for investigating & executing different Civil Engineering applications through case study presentation.	-

**25HS101 ENGLISH LANGUAGE PROFICIENCY**  
(Common to CIVIL, CSE, EEE, ECE, ICE, MECH, AI&DS and EE-VLSI)

**3 1 0 4**

**VOCABULARY:** Etymology-Prefixes and suffixes-Synonyms-Antonyms-Guessing meanings from context-Word formation- Single-word substitutes- Different forms of a word- Phrasal verbs-Collocations. **(9+3)**

**LISTENING AND SPEAKING:** Understanding listening – Listening techniques - Introducing oneself and others –Seeking and sharing information– Description-Conversation skills– Extempore speaking– Speech practice in varied formal contexts. **(9+3)**

**GRAMMAR:** Wh-questions – Yes/no questions– Parts of speech – Articles– Prepositions– Gerunds–Conjunctions-Degrees of comparison– Tenses– Modal verbs – Adverbs - Direct and indirect questions. **(9+3)**

**READING:** Reading strategies: Skimming and scanning, predicting– Reading comprehension: techniques –Practice reading. **(9+3)**

**WRITING:** Discourse markers – Dialogue writing - Completing sentences – Jumbled sentences – Paragraph writing –Writing compare & contrast paragraphs – Letter writing. **(9+3)**

**Total L: 45 + T: 15 = 60 periods**

**TEXTBOOKS:**

1. K.N. Shoba and Lourdes Joavani Rayen ‘*Communicative English*’ Cambridge University press, Cambridge, 2021.
2. Raymond Murphy ‘*Intermediate English Grammar*’ Cambridge University Press, New Delhi, 2020.
3. Dr M. Sambaiah ‘*Technical English an integrated text book*’. Wiley India Pvt. Ltd., 2025.

**REFERENCES:**

1. Raymond Murphy ‘*English Grammar in Use*’ Cambridge University Press, New Delhi 2020.
2. N.P. Sudharshana and C. Savitha ‘*English for Engineers*’ Cambridge University Press, New York, 2018.
3. Helen Naylor with Raymond Murphy ‘*Essential English Grammar*’ Cambridge University Press, New Delhi, 2019.

**COURSE OUTCOMES**

At the end of the course, students will be able to:		Bloom’s Level
<b>CO1</b>	Demonstrate the ability to recognize and use a wide range of vocabulary and key grammatical structures accurately, while developing inferential reading skills to comprehend, interpret, and analyze written texts across diverse contexts.	<b>K2</b>
<b>CO2</b>	Organize their ideas logically in essay writing, develop paragraphs with clear topic sentences and adapt their letter-writing skills to various real-world scenarios.	<b>K3</b>
<b>CO3</b>	Develop and demonstrate clear and confident speaking skills in formal and informal contexts.	

**25HS102 தமிழர் மரபு**  
(Common to CIVIL, CSE, EEE, ECE, ICE, MECH, AI&DS and EE-VLSI)

1 0 0 1

**மொழி மற்றும் இலக்கியம்:** இந்திய மொழிக் குடும்பங்கள் – திராவிட மொழிகள் – தமிழ் ஒரு செம்மொழி – தமிழ் செவ்விலக்கியங்கள் – சங்க இலக்கியத்தின் சமயச்சார் பற்ற தன்மை – சங்க இலக்கியத்தில் பகிர்தல் அறம் – திருக்குறளில் மேலாண்மைக் கருத்துக்கள் – தமிழ்க் காப்பியங்கள், தமிழகத்தில் சமண பௌத்த சமயங்களின் தாக்கம் – பக்தி இலக்கியம், ஆழ்வார்கள் மற்றும் நாயன்மார்கள் – சிற்றிலக்கியங்கள் – தமிழில் நவீன இலக்கியத்தின் வளர்ச்சி – தமிழ் இலக்கிய வளர்ச்சியில் பாரதியார் மற்றும் பாரதிதாசன் ஆகியோரின் பங்களிப்பு. (3)

**மரபு – பாறை ஓவியங்கள் முதல் நவீன ஓவியங்கள் வரை – சிற்பக்கலை:** நடுகல் முதல் நவீன சிற்பங்கள் வரை – ஐம்பொன் சிலைகள் – பழங்குடியினர் மற்றும் அவர்கள் தயாரிக்கும் கைவினைப் பொருட்கள், பொம்மைகள் – தேர் செய்யும் கலை – சுடுமண் சிற்பங்கள் – நாட்டுப்புறத் தெய்வங்கள் – குமரி முனையில் திருவள்ளூர் சிலை – இசைக்கருவிகள் – மிருதங்கம், பறை, வீணை, யாழ், நாதஸ்வரம் – தமிழர்களின் சமூக பொருளாதார வாழ்வியல் கோவில்களின் பங்கு. (3)

**நாட்டுப்புறக்கலைகள் மற்றும் வீரவிளையாட்டுகள்:** தெருக்கூத்து, கரகாட்டம், வில்லுப் பாட்டு, கணியான் கூத்து, ஓயிலாட்டம், தோல்பாவைக் கூத்து, சிலம்பாட்டம், வளரி, புலியாட்டம், தமிழர்களின் விளையாட்டுகள். (3)

**தமிழர்களின் திணைக்கோட்பாடுகள்:** தமிழகத்தின் தாவரங்களும், விலங்குகளும் – தொல்காப்பியம் மற்றும் சங்க இலக்கியத்தில் அகம் மற்றும் புறக்கோட்பாடுகள் – தமிழர்கள் போற்றிய அறக்கோட்பாடு – சங்க காலத்தில் தமிழகத்தில் எழுத்தறிவும், கல்வியும் – சங்க கால நகரங்களும் துறைமுகங்களும் – சங்க காலத்தில் ஏற்றுமதி மற்றும் இறக்குமதி – கடல்கடந்தநாடுகளில் சோழர்களின் வெற்றி. (3)

**இந்திய தேசிய இயக்கம் மற்றும் இந்திய பண்பாட்டிற்குத் தமிழர்களின் பங்களிப்பு:** இந்திய விடுதலைப் போரில் தமிழர்களின் பங்கு – இந்தியாவின் பிறப் பகுதிகளில் தமிழ்ப் பண்பாட்டின் தாக்கம் – சுயமரியாதை இயக்கம் – இந்திய மருத்துவத்தில், சித்த மருத்துவத்தின் பங்கு – கல்வெட்டுகள், கையெழுத்துப் படிக்கல்கள் – தமிழ்ப் புத்தகங்களின் அச்ச வரலாறு. (3)

**Total L: 15 periods**

**25HS102 HERITAGE OF TAMILS**  
(Common to CIVIL, CSE, EEE, ECE, ICE, MECH, AI&DS and EE-VLSI)

**1 0 0 1**

**LANGUAGE AND LITERATURE:** Language Families in India - Dravidian Languages – Tamil as a Classical Language - Classical Literature in Tamil – Secular Nature of Sangam Literature – Distributive Justice in Sangam Literature - Management Principles in Thirukural - Tamil Epics and Impact of Buddhism & Jainism in Tamil Land - Bakthi Literature Azhwars and Nayanmars - Forms of minor Poetry - Development of Modern literature in Tamil - Contribution of Bharathiyar and Bharathidhasan. (3)

**HERITAGE - ROCK ART PAINTINGS TO MODERN ART – SCULPTURE:** Hero stone to modern sculpture - Bronze icons - Tribes and their handicrafts - Art of temple car making - Massive Terracotta sculptures, Village deities, Thiruvalluvar Statue at Kanyakumari, Making of musical instruments - Mridhangam, Parai, Veenai, Yazh and Nadhaswaram - Role of Temples in Social and Economic Life of Tamils. (3)

**FOLK AND MARTIAL ARTS:** Therukoothu, Karagattam, Villu Pattu, Kaniyan Koothu, Oyillattam, Leather puppetry, Silambattam, Valari, Tiger dance - Sports and Games of Tamils. (3)

**THINAI CONCEPT OF TAMILS:** Flora and Fauna of Tamils & Aham and Puram Concept from Tholkappiyam and Sangam Literature - Aram Concept of Tamils - Education and Literacy during Sangam Age - Ancient Cities and Ports of Sangam Age - Export and Import during Sangam Age - Overseas Conquest of Cholas. (3)

**CONTRIBUTION OF TAMILS TO INDIAN NATIONAL MOVEMENT AND INDIAN CULTURE:** Contribution of Tamils to Indian Freedom Struggle - The Cultural Influence of Tamils over the other parts of India – Self-Respect Movement - Role of Siddha Medicine in Indigenous Systems of Medicine – Inscriptions & Manuscripts – Print History of Tamil Books. (3)

**Total L: 15 periods**

**TEXT BOOK**

1. V Priyadharshini ‘*தமிழர் மரபு (Heritage of Tamils)*’. VK publications, Sivakasi.

**REFERENCE BOOKS**

1. கே .கே .பிள்ளை ‘*தமிழக வரலாறு*’.- மக்களும் பண்பாடும், தமிழ்நாடு பாடநூல் மற்றும் கல்வியியல் பணிகள் கழகம் .
2. முனைவர் இல.சுந்தரம் ‘*கணினித்தமிழ்*’. விகடன் பிரசுரம்
3. கீழடி - வைகை நதிக்கரையில் சங்ககால நகர நாகரிகம், தொல்லியல் துறை வெளியீடு
4. பொருநை – ‘*ஆற்றங்கரை நாகரிகம்*’. தொல்லியல் துறை வெளியீடு
5. Dr. K.K. Pillay ‘*Social Life of Tamils*’. A joint publication of TNTB, ESC and RMRL
6. Dr. S. Singaravelu ‘*Social Life of the Tamils*’. – The Classical Period, International Institute of Tamil Studies.
7. Dr. S.V. Subramanian and Dr. K.D. Thirunavukkarasu ‘*Historical Heritage of the Tamils*’. International Institute of Tamil Studies
8. Dr. M. Valarmathi ‘*The Contributions of the Tamils to Indian Culture*’. International Institute of Tamil Studies.

9. Keeladi – ‘Sangam City Civilization on the banks of river Vaigai’. Department of Archaeology, Tamilnadu Text Book and Educational Services Corporation, Tamilnadu
10. Dr. K.K. Pillay ‘Studies in the History of India with Special Reference to Tamilnadu
11. Porunai Civilization’. Department of Archaeology, Tamil Nadu Text Book and Educational Services Corporation, Tamilnadu
12. R. Balakrishnan ‘Journey of Civilization Indus to Vaigai RMRL’. Tamilnadu

### COURSE OUTCOMES

At the end of the course, students will be able to:		Bloom’s Level
CO1	Describe the Dravidian language family, outline the features of Tamil classical literature, and explain the development of Tamil art, sculpture, and temple-related traditions in a historical context.	K2
CO2	Demonstrate the cultural relevance of Tamil folk and martial arts, apply the concepts of Sangam landscape classification to social contexts, and relate Tamil contributions to India’s freedom struggle, cultural legacy, and Siddha medicine.	K3

### 25BS111 BASIC SCIENCES LABORATORY I

0 0 4 2

#### Physics (Any eight experiments)

- Determination of rigidity modulus of a wire and moment of inertia of irregular objects – Torsional Pendulum.
- Determination of Young’s modulus of the material - Non-Uniform Bending.
- Measurement of hall coefficient of a semiconductor using Hall Effect setup.
- Determination of electrical resistivity of a given material using four probe setups.
- Determination of wavelength of laser using diffraction grating -LASER.
- Determination of Thickness of a thin wire – Air wedge method.
- Study of I-V characteristics of solar cells and determination of its efficiency.
- Determination of velocity of sound and compressibility of liquid - Ultrasonic Interferometer.
- Determination of Planck’s constant and work function of a metal - Photoelectric Effect.
- Determination of bandgap of a semiconductor – Post office box.
- Determination of specific heat capacity of a metal object.
- Verification of Hooke’s Law and calculating Spring Potential Energy.

#### Demonstration:

- Determination of Numerical Aperture and Acceptance angle - Optical Fiber
- Determination of vibration frequency by Melde’s Apparatus.

### COURSE OUTCOMES

At the end of the course, students will be able to:		Bloom’s Level
CO1	Relate the scientific principles, compare the experimental results with theoretical calculations, and apply graphical analysis to visualise the importance of precise measurements.	K3
CO2	Analyse the experimental result outcomes using analytical and experimental skills for various engineering materials and applications.	K4

**Chemistry (Any seven experiments)**

1. Determination of total, temporary & permanent hardness of water by EDTA method.
2. Determination of alkalinity in water sample.
3. Determination of strength of acids in a mixture of acids using conductivity meter.
4. Determination of strength of given hydrochloric acid using pH meter.
5. Electroplating of nickel /copper and determination of coulombic efficiency.
6. Estimation of combined ferric oxide and alumina in Portland cement.
7. Determination of Chloride in Hardened Portland Cement Concrete.
8. Photo-colorimetric estimation of Ferric ion in a water sample.

**Total P: 60 periods****REFERENCES:**

1. Department of Physics '*Physics Laboratory Observation*'. 2025.
2. Jerry D Wilson, A. Cecilia and Hernandez Hall '*Physics Laboratory Experiments*'. Boston, MA: Cengage Learning, 2016.
3. J Mendham, Vogel's '*Textbook of Quantitative Chemical Analysis 6th Ed*'. Pearson Education 2009.

**COURSE OUTCOMES**

<b>At the end of the course, students will be able to:</b>		<b>Bloom's Level</b>
<b>CO3</b>	Demonstrate the measurement of water quality parameters in the given water sample.	<b>K3</b>
<b>CO4</b>	Analyze the properties of materials for Engineering applications.	<b>K4</b>

**25CE111 ENGINEERING DRAWING****0 0 4 2****INTRODUCTION TO ENGINEERING GRAPHICS****(2)**

1. Introduction to Engineering Graphics.
2. Lettering and Dimensioning as per BIS.

**SCALES****(4)**

1. Construction of Diagonal and Vernier scales.

**ORTHOGRAPHIC PROJECTIONS****(30)**

1. Introduction to Orthographic Projections.
2. Drawing multiple views from pictorial views of objects.
3. Projection of points.
4. Projection of straight lines (only First angle projections) inclined to both the principal planes.
5. Projection of planes (polygonal and circular surfaces) inclined to both the principal planes by rotating object method.
6. Projection of simple solids when the axis is inclined to one of the principal planes and parallel to the other by rotating object method.

**SECTION OF SOLIDS****(8)**

1. Section of simple solids in simple vertical position when the cutting plane is inclined to one of the principal planes and perpendicular to the other and obtaining true shape of section.

**ISOMETRIC PROJECTIONS****(8)**

1. Isometric projection of simple solids in simple vertical positions.

**BUILDING DRAWING USING AUTOCAD****(8)**

1. Introduction to AutoCAD basic commands.
2. Plan and Elevation of Residential building using AutoCAD.

**Total P: 60 periods****TEXT BOOKS:**

1. N.D. Bhatt '*Engineering Drawing*' Charotar Publishing House Pvt. Ltd., 55<sup>th</sup> Edition, 2025.
2. K.C. John '*Engineering Graphics for Degree*' Prentice Hall India Publishers, 2009.
3. K.V. Natarajan, '*A Text book of Engineering Graphics*' Dhanalakshmi Publications, 34<sup>th</sup> Refined Edition, 2021.

**REFERENCES:**

1. K. Venugopal and V. Prabhu Raja '*Engineering Drawing + AutoCAD and Building Drawing*'. New Age International Publishers, 7<sup>th</sup> Edition, 2026.
2. Bureau of Indian Standards '*Engineering Drawing Practices for Schools and Colleges SP 46-2003*'. BIS, New Delhi, 2003

**COURSE OUTCOMES**

At the end of the course, students will be able to:		Bloom's Level
<b>CO1</b>	Use the BIS standards and specifications for drawing the engineering components and structures.	<b>K2</b>
<b>CO2</b>	Apply orthographic and pictorial projection principles to draw projections of points, lines, planes, solids, sectioned solids, development of lateral surfaces of regular and truncated solids, and isometric views engineering components and 3D objects.	<b>K3</b>
<b>CO3</b>	Develop models of typical engineering components and 3D objects manually and using software.	<b>K4</b>

**25GE111 DESIGN THINKING FOR INNOVATION****(Common to CIVIL, CSE, EEE, ECE, ICE, MECH, AI&DS and EE-VLSI)****0 0 2 1**

**Foundations of Design Thinking** History & Origins: Roots in Creative Problem Solving: Traces back to mid-20<sup>th</sup> century practices in architecture, engineering, and psychology. Herbert Simon's "Sciences of the Artificial" (1969): Introduced design as a way of thinking distinct from scientific inquiry. IDEO and the Rise of Human-Centered Design: Popularized design thinking as a repeatable, user-focused innovation process. Stanford school's Influence: Helped institutionalize design thinking in education and entrepreneurship.

Variations of Design Thinking Phases: IDEO's 3-Phase Model: Inspiration, Ideation, Implementation A flexible, non-linear approach emphasizing creativity and action. Stanford school's 5-Phase Model: Empathize, Define, Ideate, Prototype, Test A structured yet iterative framework centered on user empathy. Double Diamond Model (Design Council UK): Divides the process into Discover, Define, Develop, and Deliver—highlighting divergent and convergent thinking.

Related Concepts & Frameworks: Human-Centered Design (HCD): Focuses on designing solutions that deeply resonate with users' needs and contexts. Systems Thinking: Encourages

understanding the broader ecosystem and interdependencies within a problem space. Agile & Lean UX: Integrates design thinking with iterative development and minimal viable experimentation. Service Design: Applies design thinking to orchestrate holistic user experiences across touchpoints. Participatory Design: Involves stakeholders directly in the design process to ensure relevance and inclusivity.

**EMPATHIZE:** Apply Human-Centric Design Principles: Focus on designing solutions that prioritize user needs, experiences, and values throughout the process. Consult Experts: Engage with subject matter experts to gain foundational knowledge about the problem space. Competitive Analysis: Identify & studying similar products or services to identify gaps and opportunities. Stakeholder Interviews: Engaging with people who influence or are affected by the product or service. Conduct Observations: Observe users in their natural environment to understand behaviors, challenges, and interactions. Engage with Users: Use interviews, conversations, and other methods to connect with users and hear their stories. Immerse Yourself: Step into the users' context to experience their environment and challenges firsthand. Create Empathy Maps: Visualize what users say, think, feel, and do to synthesize insights. Identify User Needs and Pain Points: Extract meaningful patterns and needs from user interactions and observations. Set Aside Assumptions: Approach the research with an open mind, suspending personal biases and preconceptions. Document Insights: Capture quotes, observations, and emotional cues to inform the next stage (Define). (6)

**DEFINE:** Organize Research Findings: Review and structure the data collected during the Empathize stage. Analyze Observations: Identify patterns, themes, and insights from user interactions and behaviors. Craft a Human-Centered Problem Statement: Frame the problem from the user's perspective, focusing on their needs—not business goals. Avoid Business-Centric Framing: Refrain from defining problems based on company objectives alone (e.g., market share). Persona Development: Synthesizing research into user personas to guide design decisions. Use Empathy to Guide Definition: Ensure the problem statement reflects real user challenges and motivations. Develop Point-of-View Statements: Create concise summaries that capture who the user is, what they need, and why. Prepare for Ideation: Formulate “How Might We” questions to spark creative thinking in the next phase. (6)

**IDEATE:** Review the Problem Statement: Revisit the user-centric problem defined in the previous stage to guide ideation. Explore Multiple Perspectives: Encourage diverse viewpoints to broaden the range of potential solutions. Use Ideation Techniques: Apply methods like Brainstorming, Brain writing, SCAMPER, and Worst Possible Idea to spark creativity. Encourage Free Thinking: Create a judgment-free space to generate as many ideas as possible without filtering. Expand the Problem Space Push boundaries and explore unconventional or extreme ideas to uncover hidden opportunities. Refine and Select Ideas: Use evaluation techniques to identify promising concepts that address user needs effectively. Prepare for Prototyping: Choose ideas that are feasible and impactful to develop into tangible prototypes in the next stage. (6)

**PROTOTYPE:** Build Low-Cost Prototypes: Create simple, scaled-down versions of the product or its features to explore ideas. Experiment with Solutions: Implement different solutions from the Ideate stage into prototypes for testing. Test Internally and Externally: Share prototypes with team members, other departments, or a small group of users. Observe User Interactions: Watch how users engage with the prototypes to uncover usability issues and insights. Evaluate and Iterate: Accept, refine, or discard prototypes based on user feedback and performance. Identify Limitations: Discover constraints and challenges in the proposed

solutions through hands-on testing. Gain Deeper User Understanding: Learn how users think, feel, and behave when interacting with the product. (6)

**TEST:** Conduct Rigorous Testing: Evaluate the complete product using the most promising prototypes. Observe Real User Interactions: Study how users behave, think, and feel while using the product. Gather Feedback and Insights: Collect qualitative and quantitative data to assess usability and effectiveness. Identify Remaining Issues: Detect limitations, pain points, and areas for improvement. Refine and Iterate: Use test results to improve the product and revisit earlier stages if needed. Redefine Problems if Necessary: Reframe or adjust problem statements based on new insights. Enhance Understanding of Users: Deepen empathy and knowledge of user needs through real-world testing. (6)

**Design Thinking & Customer Centricity:** A human-centered approach that blends empathy and innovation to create solutions that truly resonate with customer needs. Practical Examples of Customer Challenges: Real-world scenarios where customers face friction, unmet needs, or emotional disconnects in their product or service journey. Use of Design Thinking to Enhance Customer Experience: Applying iterative problem-solving and user insights to craft experiences that are intuitive, delightful, and deeply relevant. Parameters of Product Experience: Key dimensions like usability, accessibility, emotional impact, and consistency that shape how customers perceive and interact with a product. Alignment of Customer Expectations with Product Design: Ensuring that every design decision reflects what customers value, expect, and aspire to achieve through the product.

**Total L: 30 periods**

### TEXT BOOKS

1. T. Brown '*Change by Design*'. Harper Business, 2009.
2. J. Liedtka and T. Ogilvie '*Designing for Growth*'. Columbia Business School Publishing, 2011.

### REFERENCES

1. 1.T. Kelley and D. Kelley '*Creative Confidence*'. Crown Business, 2013.
2. Stanford d. School resources: <https://dschool.stanford.edu/>
3. <https://apphaus.sap.com/toolkit/methods#design-thinking>

### COURSE OUTCOMES

At the end of the course, students will be able to:		Bloom's Level
CO1	Apply empathy-driven research to understand user needs.	K3
CO2	Frame actionable problem statements and generate creative ideas.	K5
CO3	Develop and test prototypes to refine innovative solutions to the real-world problems.	K4

**Laboratory Courses (CA: 60% + ESE: 40% Total: 100 marks)**

<b>Evaluation Scheme</b>	
<b>Internal</b>	<b>Marks</b>
• Empathize – Problem Identification & Validation	20
• Define	10
• Ideate	10
• Prototype	10
• Testing, Validation and Re-iteration	10
<b>External</b>	
• Documentation	10
• Presentation	15
• Viva	15
<b>Total</b>	<b>100</b>

**25GEM01 INDUCTION PROGRAMME**  
(Common to CIVIL, CSE, EEE, ECE, ICE, MECH, AI&DS and EE-VLSI)

All students shall undergo an induction programme at the beginning of the first semester for a duration of three weeks as per the guidelines of All India Council for Technical Education (AICTE). A student completing the induction programme will be awarded a completed grade in the grade sheet, and only the students who complete the induction programme shall be considered as eligible for award of degree subject to satisfying other conditions. A student who does not complete the induction programme in the first semester shall redo the same in the subsequent semester.

**25MA201 COMPLEX VARIABLES AND TRANSFORMS**  
(Common to CIVIL, EEE, ECE, ICE, MECH and EE-VLSI)

**3 1 0 4**

**COMPLEX DIFFERENTIATION:** Derivative, analytic function, Cauchy-Riemann equations, Laplace's equation, linear fractional transformations. **(9+3)**

**COMPLEX INTEGRATION:** Cauchy's integral theorem, Cauchy's integral formula, derivatives of analytic functions, Laurent series, singularities and zeros, residue integration method (Residue integration of complex integrals only). **(9+3)**

**LAPLACE TRANSFORMS:** Laplace transform, linearity, first shifting theorem, transforms of derivatives and integrals, unit step function, second shifting theorem, Dirac's delta function, periodic functions, differentiation and integration of transforms, solving ODEs with constant coefficients and initial value problems. **(9+3)**

**FOURIER ANALYSIS:** Fourier series – arbitrary period, even and odd functions, half range expansions. Fourier transforms, Fourier cosine and sine transforms. (9+3)

**PARTIAL DIFFERENTIAL EQUATIONS:** Basic concepts of PDEs, wave equation, heat equation, steady state two-dimensional heat problems, solution by Fourier series. (9+3)

**Total L: 45 + T: 15 = 60 periods**

### TEXT BOOKS

1. Erwin Kreyszig '*Advanced Engineering Mathematics*'. Wiley India, New Delhi, 2018.
2. G.Z. Dennis '*Advanced Engineering Mathematics*'. Jones and Bartlett Pvt Ltd, New Delhi, 2017.

### REFERENCES

1. G.Z. Dennis and D.S. Patrick '*A first course in Complex Analysis with Applications*'. Jones and Bartlett Pvt Ltd, New Delhi, 2015.
2. C.R. Wylie and L.C. Barret '*Advanced Engineering Mathematics*'. Tata McGraw-Hill, New Delhi, 2019.
3. Peter V.O Neil '*Advanced Engineering Mathematics*'. Cengage, New Delhi, 2018.
4. G.D. Dean, '*Advanced Engineering Mathematics with MATLAB*'. CRC Press, USA, 2017.

### COURSE OUTCOMES

At the end of the course, students will be able to:		Bloom's Level
CO1	Explain the concepts related to Complex Variables, Laplace Transforms, Fourier Analysis and Partial Differential Equations.	K2
CO2	Apply the techniques of Complex Variables, Laplace Transforms, Fourier Analysis and Partial Differential Equations to solve engineering problems.	K3
CO3	Analyze the solutions of engineering problems employing Complex Variables, Laplace Transforms, Fourier Analysis and Partial Differential Equations.	K4
CO4	Use modern tools to solve engineering problems with the help of Complex Variables, Laplace Transforms, Fourier Analysis and Partial Differential Equations.	-

## 25CE201 ENGINEERING STATICS AND DYNAMICS

**3 1 0 4**

**STATICS OF PARTICLES & BODIES:** Forces – Systems of forces – Concurrent forces in plane and space-Resultant equilibrium of a particle-free body diagram- rigid bodies under plane forces – moment of a force – moment of a couple – equivalent systems of coplanar forces – equilibrium of rigid bodies - types of supports - types of beams - support reactions. (10+4)

**TRUSS ANALYSIS AND FRICTION:** Analysis of trusses by method of joints and method of sections –Friction – frictional force – limiting friction – laws of static friction – coefficient of friction – angle of repose – single bodies on horizontal and inclined planes - connected bodies on horizontal and inclined planes - ladder friction - wedge friction. (9+2)

**CENTROID, AND MOMENT OF INERTIA:** Centroids of areas – centroids of simple

geometric shapes – centroids of composite areas – moment of inertia – perpendicular axis theorem – parallel axis theorem – moment of inertia of simple geometric sections – moment of inertia of composite sections – polar axis – polar moment of inertia – radius of gyration – mass moment of inertia of simple solids. (8+3)

**KINEMATICS OF PARTICLES:** Types of motion – Motion curves – Rectilinear motion – time dependent motion – uniformly accelerated motion –projectile motion. (8+3)

**KINETICS OF PARTICLES:** Rectilinear motion – Newton’s II law – D’Alembert’s principle – Energy – potential energy – kinetic energy – conservation of energy – Work done by a force – work energy method – conservation of momentum – impulse momentum principle – Impact-Direct central impact-oblique central impact. (10+3)

**Total L: 45 + T: 15 = 60 periods**

### TEXT BOOKS

1. F.P. Beer and E.R Johnson ‘*Vector Mechanics for Engineers – Statics and Dynamics*’. Tata McGraw Hill, New Delhi, 2016
2. S. Rajasekaran and G. Sankarasubramanian ‘*Engineering Mechanics – Statics and Dynamics*’. Vikas Publishing House, New Delhi, 2006

### REFERENCES

1. R.K. Bansal ‘*Engineering Mechanics*, Laxmi Publications., New Delhi, 2008
2. S.S. Bhavikatti ‘*A Text book of Engineering Mechanics*’. New Age International Pvt Ltd., New Delhi, 2008.

### COURSE OUTCOMES

At the end of the course, students will be able to:		Bloom’s Level
CO1	Elaborate the fundamental concepts and principles of the science of mechanics	K2
CO2	Compute the resultant of the coplanar forces, reactive forces under static & dynamic equilibrium conditions and moments of inertia of different cross sections	K3
CO3	Compare the geometrical and structural properties of the members under different practical conditions and propose the suitable support conditions & cross sections.	K4

## 25PH201 APPLIED PHYSICS

**3 0 0 3**

**ARCHITECTURAL ACOUSTICS:** Concepts of sound waves-propagation of sound waves-intensity. Reverberation: Reverberation Time-Salient features for an acoustically good room. Absorption- Coefficient of absorption-Absorption materials. Sabine’s Formula-Fourier spectrum. Structure of the building for an acoustically good auditorium, school, concert hall. Main parameters that affect experience—reverberation time (large spaces), early reflections, standing wave resonances (small spaces). Noise - Indoor and extraneous. Noise control methods — absorptivity, diffusers, low frequency traps. Principle of active noise cancellation. (10)

**LIGHTING AND PHOTOMETRY:** Concepts of illumination - Irradiance-brightness - Photometer - Intensity of illumination. Inverse square law and cosine law of illumination. Day

lighting. Artificial lighting. Lighting in building. Design strategies using day light. Colour perception. (8)

**THERMAL PROPERTIES:** Concepts of heat- modes of propagation of heat-conduction of heat through compound medium, radial and cylindrical flow. Thermal Conductivity-Lee's method for bad conductors- Forbes method for good conductors. Thermal comfort factors- Design for minimal heat loss by conduction, radiation. Forced convection-Air filters- air showers- Greenhouse effect. (10)

**ENERGY EFFICIENT BUILDINGS:** Electrical energy - Principle of solar cells. Manufacture of solar panels. Economic considerations in using solar power. Principle of solar collectors. Factors affecting efficiency. Building envelope - economy of air conditioning and refrigeration. Principle of Passive solar buildings. (9)

**MODERN BUILDING MATERIALS:** Composites-Sandwich and laminar panels- Pultrusion - Self Cleaning Glass panes. Metallic Glass- Shape memory alloys. Seismic Sensor materials. Bio mimetic materials. (8)

**Total L: 45 periods**

### TEXT BOOKS

1. M.N. Avadhanulu, P. G. Kshirsagar and T. V. S. Arun Murthy 'A Textbook of Engineering Physics'. S. Chand and Co., 11<sup>th</sup> Edition, 2024.
2. R.K. Gaur and S.L. Gupta 'Engineering Physics' Dhanpat Rai Publications (P) Ltd., 8<sup>th</sup> edition, New Delhi 2022.
3. William D Callister Jr., 'Materials Science and Engineering – An Introduction, 10<sup>th</sup> edition'. John Wiley and Sons, 2020.

### REFERENCES

1. Krishnan Kumar Chawla 'Composite Materials Science and Engineering'. Springer, 2019.
2. Richard Wolfson 'Essential University Physics'. Pearson Education, Singapore, 2021.
3. R.O. Davis, and A.P.S. Selvadurai 'Elasticity and Geomechanics'. Cambridge University Press, 2014.
4. Paul M. Fishbane, Stephen, Gasiorowicz and Stephen T. Thornton 'Physics for Scientists and Engineers'. Extended Version 3<sup>rd</sup> Edition, 2014.

### COURSE OUTCOMES

At the end of the course, students will be able to:		Bloom's Level
CO1	Explain the principles of acoustics, lighting, thermal properties, energy efficiency, and modern building materials, emphasizing their role in building design and construction.	K2
CO2	Apply mathematical and analytical methods to calculate sound intensity, thermal conductivity, and lighting levels in building structures using standard design principles.	K3
CO3	Analyze the impact of building materials, lighting design, and thermal management strategies on energy consumption, indoor comfort, and acoustic quality in modern constructions.	K4
CO4	Prepare a report or presentation on sustainable building design, focusing on the integration of acoustic treatments, thermal management, and energy-efficient systems in contemporary architecture.	-

**25CY201 CHEMISTRY OF BUILDING MATERIALS II****3 0 0 3**

**METALLIC MATERIALS IN CONSTRUCTION:** Ferrous metals: Iron ores – Manufacturing, Properties and uses of cast iron, wrought iron, pig Iron. Manufacturing, properties and uses of steel – mild steel, stainless steel – characteristics, high carbon steel. Non – ferrous: metals- Al, Cu, Zn, Alloys of Al, Cu, Mg, Ni – production, Properties and uses. Anodizing of aluminum, applications. (9)

**GLASSES, INSULATING AND COMPOSITE MATERIALS:** Glass – types, application in construction. Types of insulating materials – air spaces, aerated concrete, gypsum, expanded blast furnace slag, sprayed asbestos, vermiculate, coconut fibers, cork board, cellulose, fibre glass. Sound and electrical insulating materials. Classification and constituents of composites, fiber – reinforced composites and failure – particle reinforced composites, particulate composites, structural composites, advanced composite. (9)

**GYPSUM, BITUMINOUS MATERIALS AND MIXTURES:** Gypsum-Setting and hardening of gypsum- classification and manufacturing of gypsum-application- wall plasters. Bitumen, tar, pitches and asphalts – petroleum asphalts – asphalt cement, cut back asphalts, emulsified asphalts, air blown asphalts. Properties of asphalts – adhesion, specific gravity, durability, rate of curing, ageing and hardening and resistance to reaction with water. (9)

**LOW-COST BUILDING MATERIALS:** Industrial wastes: Fly ash, blast furnace slag, copper slag, red mud aggregate, steel making slag, ferro alloy slag, mine wastes, mica scrap, gypsum mines waste, Iron and gold tailings- coal washery waste and copper mine tailing Agricultural wastes - Rice husk, saw mill dust, jute stalk, sisal fibres, coconut wastes. (9)

**INNOVATIVE BUILDING MATERIALS:** Fly ash bricks, Soil – cement blocks, soil-lime blocks, mud blocks, calcium silicate bricks, Fal-G-cement, corrugated roofing sheets from coir waste. Red-mud, jute fibre polymer composite, expanded polystyrene composite, glass reinforced gypsum-medium fibre boards. (9)

**Total L: 45 Periods****TEXT BOOKS**

1. J. Neil and R.K. Dhir '*Civil Engineering Materials*'. Brijbasi Art Press Ltd., Noida, 2007.
2. U.K. Shrivastava '*Building Materials Technology*'. Galgotia Publications Pvt Ltd, New Delhi., 1st Ed, 2012.

**REFERENCES**

1. P.D. Kulkarni, R. Subramanian, P.S. Galion, A. Juneja, V.P. Puri and S.K. Likhi '*Civil Engineering Materials*'. Tata McGraw Hill., New Delhi, 1997.
2. S.K. Duggal '*Building Materials*'. New Age International Publishers., New Delhi, 5<sup>th</sup> Edition, 2019.
3. R.K. Rajput '*Engineering Materials – Construction Materials*'. S Chand and Company, New Delhi, Revised 5th Ed., 2016.

**COURSE OUTCOMES**

At the end of the course, students will be able to:		Bloom's Level
CO1	Learn the chemistry of Materials for Civil Engineering	K2
CO2	Choose the materials for building and construction purpose	K3
CO3	Correlate the properties of materials for efficient construction	K4

**25HS201 தமிழரும் தொழில்நுட்பமும்**  
(Common to CIVIL, CSE, EEE, ECE, ICE, MECH, AI&DS and EE-VLSI)

1 0 0 1

**நெசவு மற்றும் பாணைத் தொழில்நுட்பம்:** சங்க காலத்தில் நெசவுத் தொழில் – பாணைத் தொழில்நுட்பம் – கருப்பு சிவப்பு பாண்டங்கள் – பாண்டங்களில் கீறல் குறியீடுகள். (3)

**வடிவமைப்பு மற்றும் கட்டிடத் தொழில்நுட்பம்:** சங்க காலத்தில் வடிவமைப்பு மற்றும் கட்டுமானங்கள், சங்க காலத்தில் வீட்டுப் பொருட்களில் வடிவமைப்பு – சங்க காலத்தில் கட்டுமான பொருட்களும் நடுகல்லும் – சிலப்பதிகாரத்தில் மேடை அமைப்பு பற்றிய விவரங்கள் – மாமல்லபுரச் சிற்பங்களும், கோவில்களும் – சோழர் காலத்துப் பெருங்கோயில்கள் மற்றும் பிற வழிபாட்டுத் தலங்கள் – நாயக்கர் காலக் கோயில்கள் – மாதிரி கட்டமைப்புகள் பற்றி அறிதல், மதுரை மீனாட்சி அம்மன் ஆலயம் மற்றும் திருமலை நாயக்கர் மஹால் – செட்டிநாட்டு வீடுகள் – பிரிட்டிஷ் காலத்தில் சென்னையில் இந்தோ-சாரோசெனிக் கட்டிடக் கலை. (3)

**உற்பத்தித் தொழில் நுட்பம்:** கப்பல் கட்டும் கலை – உலோகவியல் – இரும்புத் தொழிற்சாலை – இரும்பை உருக்குதல், எஃகு – வரலாற்றுச் சான்றுகளாக செம்பு மற்றும் தங்க நாணயங்கள் – நாணயங்கள் அச்சடித்தல்- மணி உருவாக்கும் தொழிற்சாலைகள் – கல்மணிகள், கண்ணாடி மணிகள் – சுடுமண் மணிகள் – சங்கு மணிகள் – எலும்புத் துண்டுகள் – தொல்லியல் சான்றுகள் – சிலப்பதிகாரத்தில் மணிகளின் வகைகள். (3)

**வேளாண்மை மற்றும் நீர்ப்பாசனத் தொழில் நுட்பம்:** அணை, ஏரி, குளங்கள், மதகு – சோழர்காலக் குழுழித் தூம்பின் முக்கியத்துவம் – கால்நடை பராமரிப்பு – கால்நடைகளுக்காக வடிவமைக்கப்பட்ட கிணறுகள் – வேளாண்மை மற்றும் வேளாண்மைச் சார்ந்த செயல்பாடுகள் – கடல்சார் அறிவு – மீன்வளம் – முத்து மற்றும் முத்துக்குளித்தல் – பெருங்கடல் குறித்த பண்டைய அறிவு – அறிவுசார் சமூகம். (3)

**அறிவியல் தமிழ் மற்றும் கணினித்தமிழ்:** அறிவியல் தமிழின் வளர்ச்சி – கணினித்தமிழ் வளர்ச்சி – தமிழ் நூல்களை மின்பதிப்பு செய்தல் – தமிழ் மென்பொருட்கள் உருவாக்கம் – தமிழ் இணையக் கல்விக்கழகம் – தமிழ் மின் நூலகம் – இணையத்தில் தமிழ் அகராதிகள் – சொற்குவைத் திட்டம். (3)

**25HS201 TAMILS AND TECHNOLOGY**  
(Common to CIVIL, CSE, EEE, ECE, ICE, MECH, AI&DS and EE-VLSI)

**1 0 0 1**

**WEAVING AND CERAMIC TECHNOLOGY:** Weaving Industry during Sangam Age – Ceramic technology – Black and Red Ware Potteries (BRW) – Graffiti on Potteries. (3)

**DESIGN AND CONSTRUCTION TECHNOLOGY:** Designing and Structural construction House & Designs in household materials during Sangam Age - Building materials and Hero stones of Sangam age – Details of Stage Constructions in Silappathikaram - Sculptures and Temples of Mamallapuram - Great Temples of Cholas and other worship places - Temples of Nayaka Period - Type study (Madurai Meenakshi Temple)- Thirumalai Nayakar Mahal - Chetti Nadu Houses, Indo - Saracenic architecture at Madras during British Period. (3)

**MANUFACTURING TECHNOLOGY:** Art of Ship Building - Metallurgical studies - Iron industry - Iron smelting, steel -Copper and gold- Coins as source of history - Minting of Coins – Beads making-industries Stone beads -Glass beads - Terracotta beads -Shell beads/ bone beads - Archeological evidences - Gem stone types described in Silappathikaram. (3)

**AGRICULTURE AND IRRIGATION TECHNOLOGY:** Dam, Tank, ponds, Sluice, Significance of Kumizhi Thoompu of Chola Period, Animal Husbandry - Wells designed for cattle use - Agriculture and Agro Processing - Knowledge of Sea - Fisheries – Pearl - Conche diving - Ancient Knowledge of Ocean - Knowledge Specific Society. (3)

**SCIENTIFIC TAMIL & TAMIL COMPUTING:** Development of Scientific Tamil - Tamil computing – Digitalization of Tamil Books – Development of Tamil Software – Tamil Virtual Academy – Tamil Digital Library – Online Tamil Dictionaries – Sorkuvai Project. (3)

**Total L: 15 periods**

**TEXT BOOK**

1. V Priyadharshini ‘தமிழரும் தொழில்நுட்பமும் (Tamil and Technology)’. VK publications, Sivakasi.

**REFERENCE BOOKS**

1. கே .கே .பிள்ளை ‘தமிழக வரலாறு - மக்களும் பண்பாடும்’. தமிழ்நாடு பாடநூல் மற்றும் கல்வியியல் பணிகள் கழகம் .
2. முனைவர் இல.சுந்தரம் ‘கணினித்தமிழ்’. விகடன் பிரசுரம்
3. ‘கீழடி - வைகை நதிக்கரையில் சங்ககால நகர நாகரிகம்’.தொல்லியல் துறை வெளியீடு
4. ‘பொருதை - ஆற்றங்கரை நாகரிகம்’. தொல்லியல் துறை வெளியீடு
5. Dr. K.K. Pillay ‘Social Life of Tamils’. A joint publication of TNTB, ESC and RMRL
6. Dr. S. Singaravelu ‘Social Life of the Tamils – The Classical Period’. International Institute of Tamil Studies.
7. Dr. S.V. Subramanian and Dr. K.D. Thirunavukkarasu ‘Historical Heritage of the Tamils’. International Institute of Tamil Studies
8. Dr. M. Valarmathi ‘The Contributions of the Tamils to Indian Culture’. International Institute of Tamil Studies.

9. 'Keeladi – Sangam City Civilization on the banks of river Vaigai'. Department of Archaeology, Tamilnadu Text Book and Educational Services Corporation, Tamilnadu
10. Dr. K.K. Pillay 'Studies in the History of India with Special Reference to Tamilnadu
11. Porunai Civilization', Department of Archaeology, Tamil Nadu Text Book and Educational Services Corporation, Tamilnadu
12. R. Balakrishnan 'Journey of Civilization Indus to Vaigai'. RMRL, Tamilnadu

### COURSE OUTCOMES

At the end of the course, students will be able to:		Bloom's Level
CO1	Identify the significance of ancient Tamil technologies in weaving, pottery, metallurgy, and architecture, with emphasis on traditional design and construction methods across historical periods.	K2
CO2	Use insights from traditional Tamil knowledge systems in agriculture, irrigation, and marine sciences, and connect the development of Tamil language to its applications in digital platforms and computing.	K3

### LANGUAGE ELECTIVES

#### 25HS211 COMMUNICATION SKILLS FOR ENGINEERS (Common to CIVIL, CSE, EEE, ECE, ICE, MECH, AI&DS and EE-VLSI)

**0 0 4 2**

**COMMUNICATION CONCEPTS:** Process of Communication – Inter and Intrapersonal Communication – Essentials for effectiveness. **(9)**

**ORAL COMMUNICATION:** Oral presentations with visual aids and Group discussions. **(16)**

**FOCUS ON SOFT SKILLS:** Etiquette – Work Place etiquette – Telephone etiquette- Body Language – Critical Reasoning and Conflict Management based on Case Studies – Group Communication- Meetings -Interview Techniques. **(14)**

**TECHNICAL WRITING:** Technical Writing Principles - Style and Mechanics - Technical Definitions – Physical, Functional and Process Descriptions – Technical Report Writing – Preparing Instructions – Interpretation of Technical Data. **(14)**

**BUSINESS CORRESPONDENCE:** Writing Emails, Preparing Resumes. **(7)**

**Total P: 60 periods**

### TEXT BOOKS

1. Course materials prepared by the Faculty, Department of English.

**REFERENCES**

1. Jeff Butterfield '*Soft Skills for Everyone*'. Cengage Learning, New Delhi, 2020.
2. Sabina Pillai and Agna Fernandez '*Soft skills and Employability Skills*'. Cambridge University Press, New Delhi, 2019.
3. Prashant Sharma '*Soft Skills Personality Development for Life Success*'. BPB Publications, New Delhi, 2021.
4. K.N. Shoba and D. Praveen Sam '*Technical English*'. Cambridge University Press, New York, 2020.

**COURSE OUTCOMES**

At the end of the course, students will be able to:		Bloom's Level
CO1	Produce clear and concise technical reports, compose professional and effective emails and develop well-structured and impactful resumes	K2
CO2	Plan, organize, and deliver engaging and informative presentations using appropriate visual aids and participate positively in group discussions	K3
CO3	Resolve disagreements constructively, embody professional conduct and a strong work ethic and apply critical thinking to generate effective solutions	

**25HS212 BASIC GERMAN**

(Common to CIVIL, CSE, EEE, ECE, ICE, MECH, AI&amp;DS and EE-VLSI)

**0 0 4 2**

**Guten Tag! - Learning:** To greet, learn numbers till 20, practice telephone numbers & e mail address, learn alphabet, speak about countries & languages; **Vocabulary:** related to the topic; **Grammar:** W – Questions, Verbs & Personal pronouns I.

**Freunde, Kollegen und ich - Learning:** To speak about hobbies, jobs, learn numbers from 20; **Vocabulary:** related to the topic; **Grammar:** Articles, Verbs & Personal pronouns II, sein & haben verbs, ja/nein Frage, singular/plural.

**In der Stadt – Learning:** To know places, buildings, question, know transport systems, understand international words; **Vocabulary:** related to the topic; **Grammar:** Definite & indefinite articles, Negotiation, Imperative with Sie.

**Guten Appetit! – Learning:** To speak about food, shop, converse; **Vocabulary:** related to the topic; **Grammar:** Sentence position, Accusative, Accusative with verbs.

**Tag für Tag and Zeit mit Freunden – Learning:** To learn time related expressions, speak about family, ask excuse, fix appointments on phone, birthdays, understand & write invitations, converse in the restaurant; **Vocabulary:** related to the topic; **Grammar:** Preposition – am, im, um, von...bis, Possessive articles, Modal verbs.

**Total L: 60 periods****TEXT BOOK:**

1. Dengler, Stefanie et al. '*Netzwerk A1.1*'. Klett-Langenscheidt Gmbh, München, 2013.

**REFERENCES:**

1. Dengler, Stefanie et al. *'Netzwerk AI'*. Klett-Langenscheidt GmbH München, 2013.
2. Sandra Evans, Angela Pude, Franz Specht- *'Menschen AI'*. –Hueber Verlag, 2012.
3. Hermann Funk, Christina Kuhn, Silke Demme *'Studio d AI'*. Goyal Publishers & Distributors Pvt. Ltd, 2009.
4. Rosa-Maria Dallapiazza, Eduard von Jan *'Til Schönherr'*. Tangram Aktuell 1 (Deutsch als Fremdsprache), Max Hueber Verlag, 2004.

**25HS213 BASIC JAPANESE****(Common to CIVIL, CSE, EEE, ECE, ICE, MECH, AI&DS and EE-VLSI)****0 0 4 2**

Orientation Session, Geographic & Socio, economic perspective to Japan, Japanese people and culture and Basic greetings and responses.

Basic script, Method of writing hiragana and katakana, and Combination sounds and simple words.

Topic marker “wa”, Desu / dewa arimasen cupolas, Interrogative particle “ka”, Grammar particles “mo”, “no”, “Introducing some one: “Kochira wa ~“ and Self introductions: Hajimemashite”

Demonstratives “Kore”, “Sore”, “Are”, Demonstrative “Kono”, “Sono”, “Ano”, Possessive noun particle “no” and Japanese apartments: Greeting your neighbor.

Place markers “Koko”, “Soko”, “Asoko”, Direction markers “Kochira”, “Sochira”, “Achira” and Japanese department stores: Asking for and buying something.

Asking for and telling the time, Particle “ni (at)” for time, kara (from) ~ made (until), Particle “to (and)”, Time periods: Days of the week, months, time of day, Verbs (Present / future and past tense) and Telephone enquiry: Asking for a phone no. And business hours.

Destination particle “e”, Particles “de (mode of transportation)” and “to (with) and Japanese train station: Asking for Fare and track no. / types of trains.

Direct object particle “o”, Particle “de (place of action)”, Verbs (“~masen ka”, “~mashou”) and “Ohanami” Cherry blossom viewing.

Particle “de (by means of)”, Particle “ni (to)”, Aemasu (give) and Moraimasu (receive) and Visiting a Japanese house.

Adjectives (“i” and “na” type), Adjectives (Positive and negative usage), Particle “ga (however, but), “Dore which?” and Leaving a room, thanking someone for hospitality.

Likes and dislikes, Potential verbs (wakarimasu and dekimasu), “Kara ( ~ because)”, Adverbs and Asking someone out over the phone.

Verbs denoting presence: “Imasu” and “arimasu”, Particle “ni (in)”, “Dare (who?)”, Adverbs (“Chikaku ni ~ “), Particle “dare mo (negative ~ no one)”, Dare ka (anyone), dare ga (who) , Nani ka (anything) , nani ga (what) - ~ya (and) ~ nado (etc.) and Asking for directions.

Counters and Counting suffixes.

Introduction to Adjectives (na and ii type), Different usages of adjectives, Comparison, Likes and dislikes and Going to a trip.

Need and desire (ga hoshii), Wanting to ... (Tabeti desu), Going for a certain purpose (mi – ni ikimasu) and Choosing from a menu.

Verb groups, I, II and III and Exercises to group verbs.

Please do (te kudasai), Present continuous tenses (te imasu), Shall I? (~ mashou ka) and Describing a natural phenomenon (It is raining).

To grant permission (~te mo ii desu), Asking for permission (~ te mo ii desu ka) and Should not do (~ te waikemasen) Describing a continuing state and Describing a habitual action.

Roleplays in Japanese.

A demonstration on usage of chopsticks and Japanese tea party.

**Total L: 60 periods**

### **TEXT BOOK**

1. ‘*Minna no nohongo*’ – Romaji ban (first 10 lessons of this book).

### **REFERENCE**

1. ‘*Minna no Nihongo I Honsatsu Roma – ji ban (Main Textbook Romanized Version)*’. International publisher – 3A Corporation, Tokyo, Indian distributor – Goyal Publishers & Distributors, New Delhi.

## **25BS211 BASIC SCIENCES LABORATORY II**

**0 0 4 2**

### **Physics (Any eight experiments)**

1. Determination of velocity of sound in liquid – Acoustic grating.
2. Determination of Young’s modulus of the material- Uniform Bending.
3. Determination of Young’s modulus of the material – Simple harmonic Oscillations of a cantilever.
4. Determination of thermal conductivity of a bad conductor using Lee’s Disc.
5. Determination of particle size of a microparticle film using LASER.
6. Determination of coefficient of viscosity of a liquid – Poiseuille’s method.
7. Study the dispersion relation and determination of cut off frequency using lattice dynamics kit.
8. Determination of wavelength of mercury spectrum using spectrometer and grating.
9. Determination of specific resistance of a given wire – Carey foster’s bridge.
10. Determination of thermal conductivity of a metallic material – Wiedemann – Franz Law.
11. Determination of force in members of Truss Bridge.
12. Validation of Newton’s laws of motion.

**Demonstration:**

1. Study the energy loss of a ferrite magnetic material specimen by B-H curve.
2. Determination of resolving power of a prism using spectrometer.

**COURSE OUTCOMES**

At the end of the course, students will be able to:		Bloom's Level
CO1	Relate the scientific principles, compare the experimental results with theoretical calculations and apply graphical analysis to visualise the importance of precise measurements.	K3
CO2	Analyse the experimental result outcomes using analytical and experimental skills for various engineering materials and applications	K4

**Chemistry (Any eight experiments)**

1. Estimation of  $\text{Ca}^{2+}$  and  $\text{Mg}^{2+}$  hardness of water by EDTA method
2. Determination of DO content of water sample by Winkler's method
3. Estimation of iron content of the given solution using potentiometer.
4. Corrosion experiment-weight loss method.
5. Estimation of percentage of calcium oxide present in Portland cement.
6. Anodizing of aluminium and determination of thickness of anodised film.
7. Determination of kinematic viscosity of lubricating oil using Redwood viscometer.
8. Construction of phase diagram of a simple eutectic system.

**Total P: 60 periods****REFERENCES:**

1. Department of Physics '*Physics Laboratory Observation*'. 2025.
2. Jerry D Wilson, A. Cecilia and Hernandez Hall '*Physics Laboratory Experiments*'. Boston, MA, Cengage Learning, 2016.
3. J Mendham, '*Vogel's Textbook of Quantitative Chemical Analysis, 6th Ed.,*'. Pearsons Education 2009.

**COURSE OUTCOMES**

At the end of the course, students will be able to:		Bloom's Level
CO3	Demonstrate the measurement of water quality parameters in the given water sample.	K3
CO4	Analyze the properties of materials for Engineering applications.	K4

**25CE211 ENGINEERING PRACTICES LABORATORY****0 0 2 1****ENGINEERING PRACTICES:**

1. Carpentry: Study of wood working tools. Exercises: Preparation of "L" and "V" Joints.
2. Plumbing: Study of tools and operations; Exercises: External thread cutting and preparation of PVC pipe joints.
3. Welding: Study of arc welding tools and equipment; Exercises: Preparation of joints using arc and TIG welding.
4. Sheet metal work and Soldering: Study of tools and operations; Exercise: Preparation of a rectangular tray.
5. Basic Machining: Study of tools and operations, Study of types of connections; Exercise: Simple turning and taper turning, Drilling - bolted connections.

**Total P: 30 periods**

**REFERENCES**

1. Department of Mechanical Engineering, ‘*Manufacturing Processes Laboratory Manual*’. PSG College of Technology, 2019.
2. W.A.J. Chapman, ‘*Workshop Technology - Part I, 4th Edition*’. CBS Publications, New Delhi, 2007.

**COURSE OUTCOMES**

At the end of the course, students will be able to:		Bloom’s Level
CO1	Apply basic manufacturing and fabrication techniques such as welding, carpentry, plumbing, turning, drilling and sheet metal work using appropriate tools and procedures to prepare simple joints and components.	K3

**25CE212 PROBLEM SOLVING USING PYTHON PROGRAMMING  
LABORATORY**

**0 0 4 2**

**List of Experiments:**

1. Identification and solving of simple real life or scientific or technical problems, and developing flow charts for the same. (Electricity Billing, Retail shop billing, Sin series, weight of a motorbike, Weight of a steel bar, compute Electrical Current in Three Phase AC Circuit, etc.)
2. Python programming using simple statements and expressions (exchange the values of two variables, circulate the values of n variables, distance between two points).
3. Scientific problems using Conditionals and Iterative loops. (Number series, Number Patterns, pyramid pattern)
4. Implementing real-time/technical applications using Lists, Tuples. (Items present in a library/Components of a car/ Materials required for construction of a building –operations of list & tuples)
5. Implementing real-time/technical applications using Sets, Dictionaries. (Language, components of an automobile, Elements of a civil structure, etc.- operations of Sets & Dictionaries)
6. Implementing programs using Functions. (Factorial, largest number in a list, area of shape)
7. Implementing programs using Strings. (reverse, palindrome, character count, replacing characters)
8. Implementing programs using written modules and Python Standard Libraries (pandas, numpy. Matplotlib, scipy)
9. Implementing real-time/technical applications using File handling. (copy from one file to another, word count, longest word)
10. Implementing real-time/technical applications using Exception handling. (divide by zero error, voter’s age validity, student mark range validation)
11. Exploring Pygame tool
12. Developing a game activity using Pygame like bouncing ball, car race etc.

**Total P: 60 periods**

**TEXT BOOKS**

1. Allen B. Downey ‘*Think Python: How to Think like a Computer Scientist 2nd Edition*’. O’Reilly Publishers, 2016.

2. Karl Beecher '*Computational Thinking: A Beginner's Guide to Problem Solving and Programming 1st Edition*'. BCS Learning and Development Limited, 2017.

## REFERENCES

1. Paul Deitel and Harvey Deitel '*Python for Programmers*'. Pearson Education, 1st Edition, 2021.
2. G Venkatesh and Madhavan Mukund '*Computational Thinking: A Primer for Programmers and Data Scientists 1st Edition*'. Notion Press, 2021.
3. John V Guttag '*Introduction to Computation and Programming Using Python: With Applications to Computational Modeling and Understanding Data*'. Third Edition, MIT Press, 2021
4. Eric Matthes '*Python Crash Course, A Hands - on Project Based Introduction to Programming 2nd Edition*'. No Starch Press, 2019.
5. <https://www.python.org/>
6. Martin C. Brown '*Python: The Complete Reference 4th Edition*'. Mc-Graw Hill, 2018.

## COURSE OUTCOMES

At the end of the course, students will be able to:		Bloom's Level
CO1	Develop algorithmic solutions to simple computational problems	K2
CO2	Develop and execute simple Python program for real time problems	K3

## 25EEEC01 WORKPLACE COMMUNICATION SKILLS

(Common to CIVIL, CSE, EEE, ECE, ICE, MECH, AI&DS and EE-VLSI)

0 0 2 0

## BUILDING COMMUNICATION SKILLS:

1. Introduction to Workplace Communication
2. Profile Building for Internships
3. English in the Workplace (Grammar & Vocabulary)
4. Professional Communication (Speaking & Writing)
5. Workplace Communication Tools
6. Career Exploration
7. Resume Update

**Total P: 30 periods**

## REFERENCES:

1. P.C. Wren and H. Martin '*High school English Grammar and Composition*'. S Chand Publishing, New Delhi, 2017.
2. Norman Lewis '*Word Power Made Easy*'. Goyal Publisher, New Delhi, 2011.

## COURSE OUTCOMES

At the end of the course, students will be able to:		Bloom's Level
CO1	Understand the importance of soft skills for employability and fine tune their writing skills – Resume writing	K3
CO2	Present with clarity and coherence while speaking in formal contexts.	K2